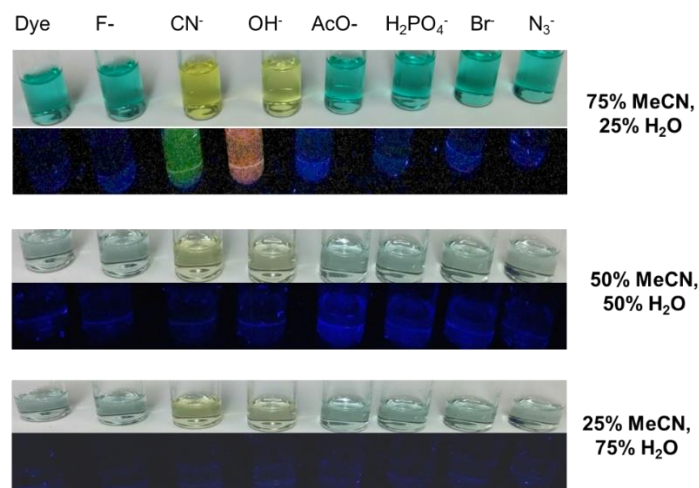
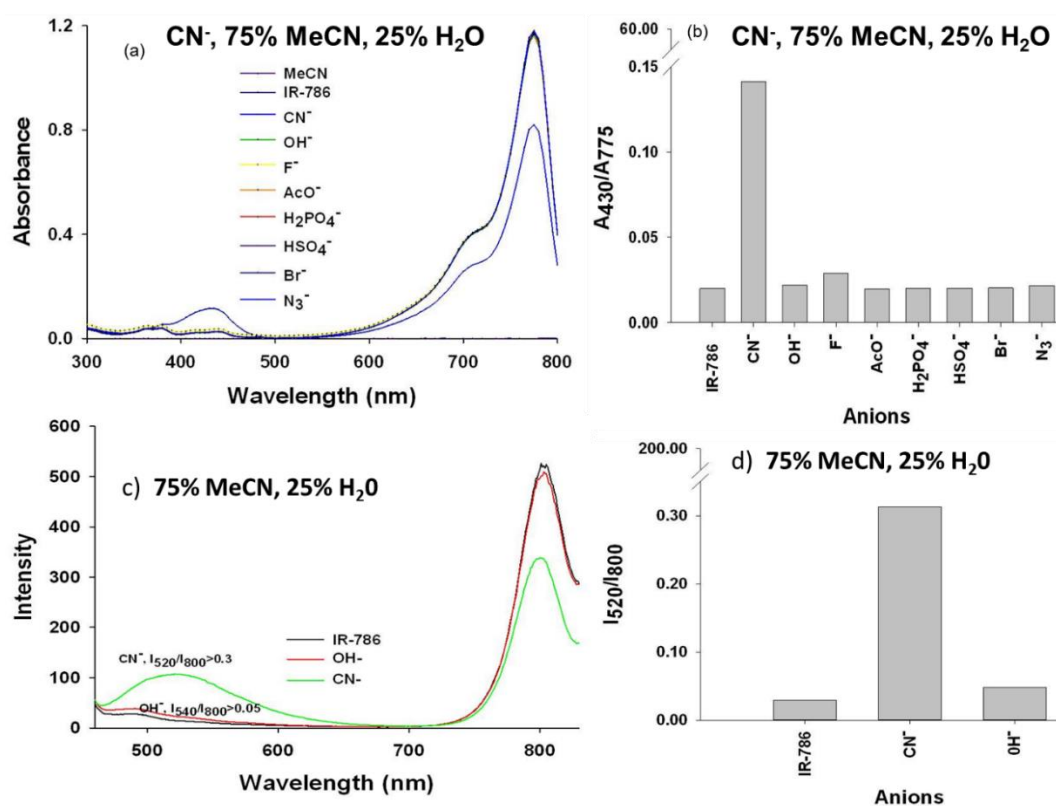


# Supplementary Materials: A Highly Selective Sensor for Cyanide in Organic Media and on Solid Surfaces

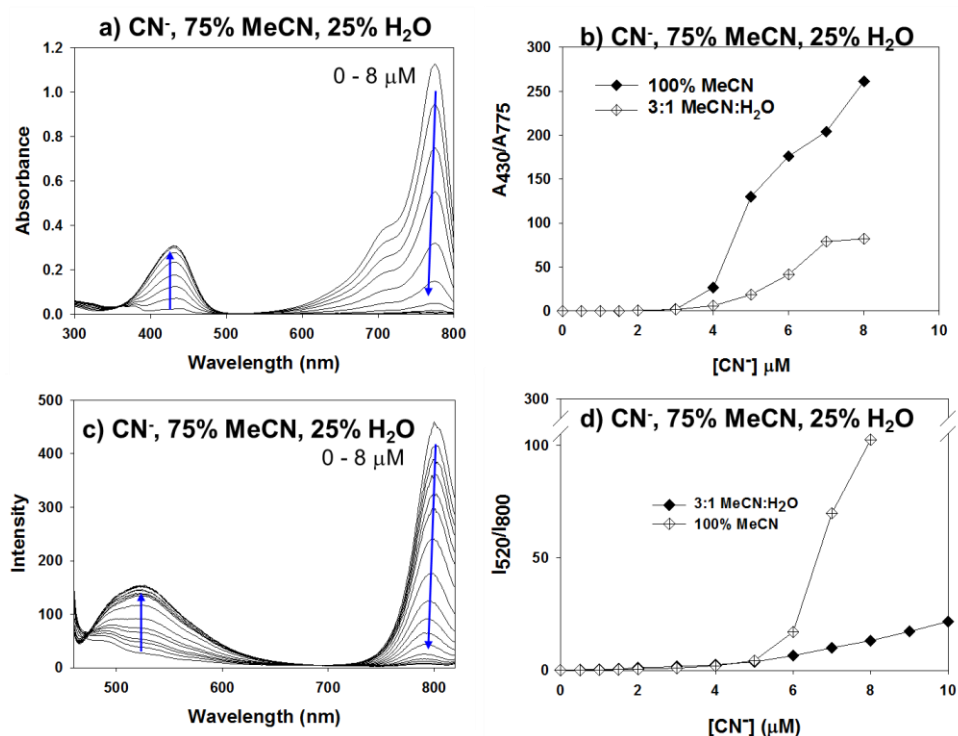
Belygona Barare, Ilknur Babahan, Yousef M. Hijji, Enock Bonyi, Solomon Tadesse and Kadir Aslan



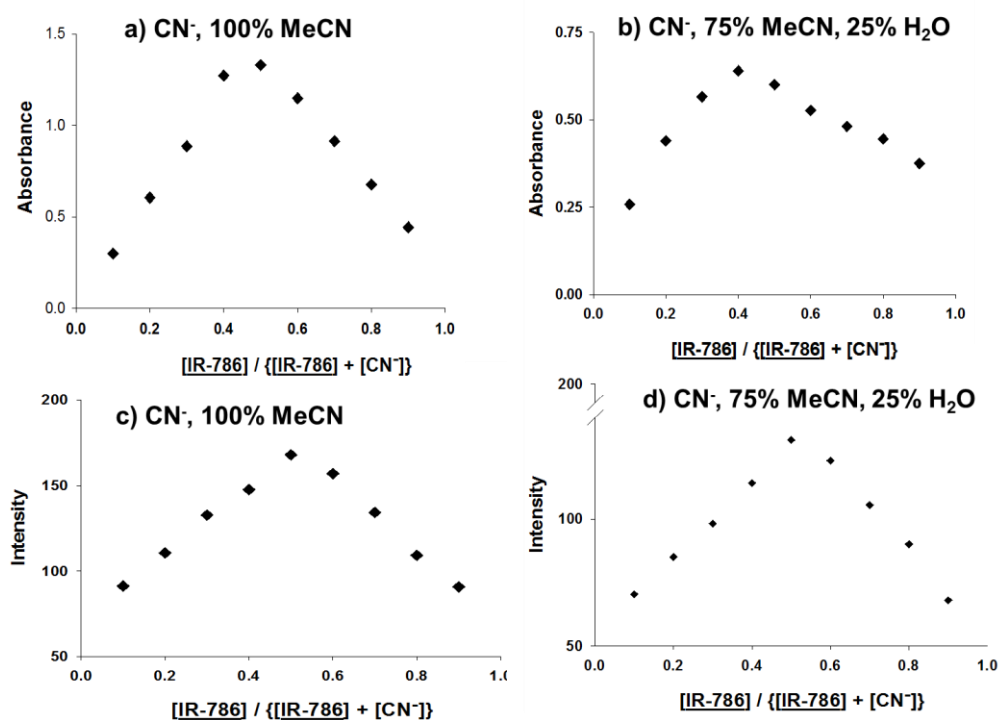
**Figure S1.** Color changes of IR-786 (50 μM) under normal light and under hand held UV lamp ( $\lambda = 365$  nm) in MeCN: water mixtures = 75%:25%, 50%:50%, 25%:75% before and after the addition of various anions.



**Figure S2.** (a) UV-vis absorption spectra and (b) absorbance ratiometric values (absorbance value at 430 divided by absorbance value at 775 nm) before and after the addition of 1.0 equivalence of various anions to IR-786 in 75%:25% MeCN:water mixture (50 μM) and (c) emission spectra and (d) emission ratiometric values (intensity value at 520 nm divided by intensity value at 800 nm) obtained from (a).

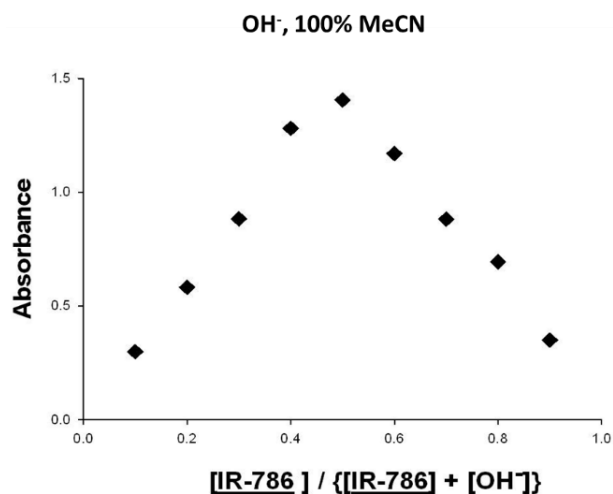


**Figure S3.** (a) UV-vis absorption spectra of IR-786 in 75%:25% MeCN:water mixture (50  $\mu\text{M}$ ) before and after the addition of  $\text{CN}^-$  (up to 8.0  $\mu\text{M}$ ) and (b) absorbance ratiometric values (absorbance value at 430 divided by absorbance value at 775 nm) obtained from (c) Fluorescence emission spectra ( $\lambda_{\text{ex}} = 430$  nm, excitation slit = 20 nm, emission slit = 20 nm), for IR-786 (5.0  $\mu\text{M}$ ) 75%:25% MeCN:water mixture before and after the addition of  $\text{OH}^-$  (up to 10  $\mu\text{M}$ ) and (d) fluorescence emission ratiometric values (intensity value at 520 nm divided by intensity value at 800 nm) obtained from (c). Arrows show the direction of increased amount of anions.

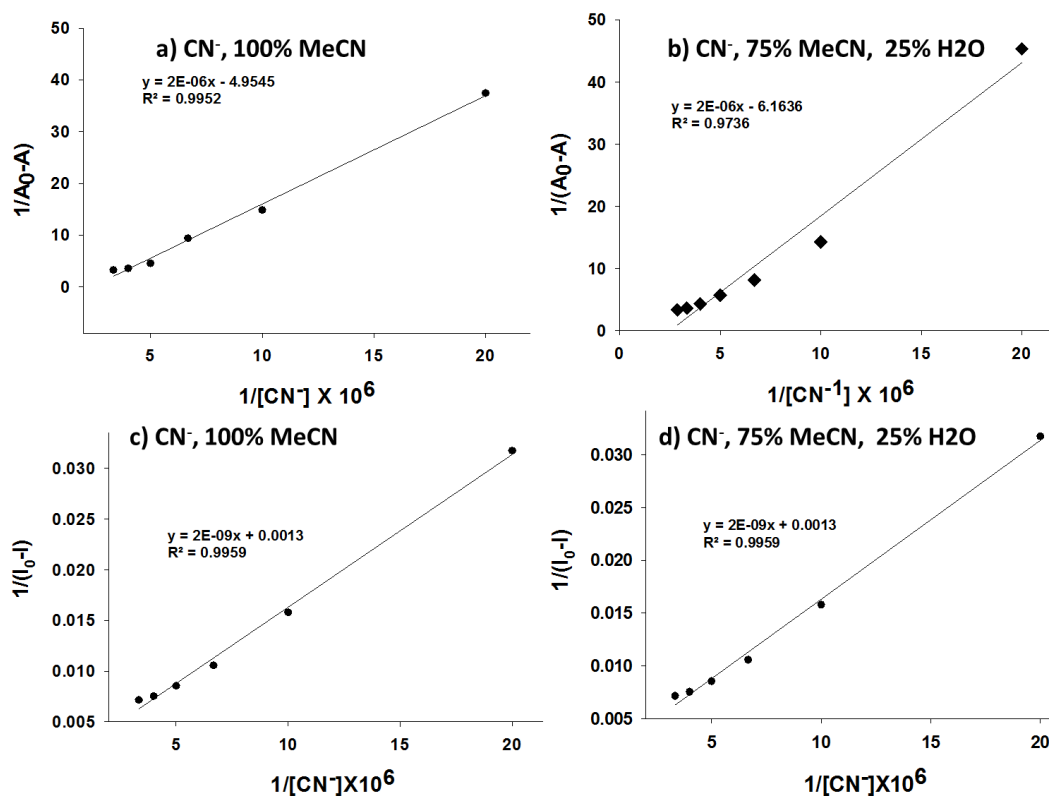


**Figure S4.** Job's plots for the determination of the binding stoichiometry between IR-786 (5.0  $\mu\text{M}$ ) and  $\text{CN}^-$  in (a) 100% MeCN (b) 75%:25% MeCN:H<sub>2</sub>O mixture based on absorbance at  $\lambda_{\text{max}} = 430$  nm

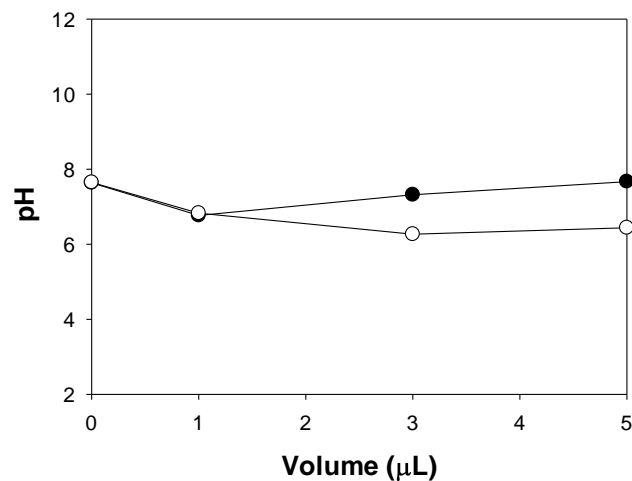
and (c) 100% MeCN (d) 75%:25% MeCN:H<sub>2</sub>O mixture based on fluorescence emission at  $\lambda_{\text{max}} = 520$  nm.



**Figure S5.** Job's plot for the determination of the binding stoichiometry between IR-786 (5.0  $\mu\text{M}$ ) and OH<sup>-</sup> in 100% MeCN based on absorbance at  $\lambda_{\text{max}} = 430$  nm.



**Figure S6.** Plots for the determination of the binding constants between IR-786 (50  $\mu\text{M}$ ) and CN<sup>-</sup> in (a) 100% MeCN (b) 75%:25% MeCN:H<sub>2</sub>O mixture based on absorbance at  $\lambda_{\text{max}} = 430$  nm and (c) 100% MeCN (d) 75%:25% MeCN:H<sub>2</sub>O mixture based on fluorescence emission at  $\lambda_{\text{max}} = 520$  nm.



**Figure S7.** Change in pH of IR-786 solution in MeCN during the titration of stock solution of  $\text{CN}^-$  and  $\text{OH}^-$  in buffer as described in the experimental section. Volume of anions correspond to 0–20  $\mu\text{M}$ .