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# Phenolic profile, nutritional potential and biological activities of wildly grown accessions of *Cucumis melo* var. *Agrestis*



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# ABSTRACT

The aim of the present research work was to investigate the variation in nutritional potential, phenolic profile, antioxidant, and free radical scavenging activities of four wildly grown accessions of Cucumis melo var. Agrestis. The accessions of Cucumis melo var. Agrestis were collected from different areas of the district Faisalabad and Layyah of Punjab province (Pakistan). Mineral profile and amino acid analysis were examined using inductively coupled plasma optical emission spectroscopy (ICP-OES) and an amino acid analyzer. High-pressure liquid chromatography (HPLC) was used for the separation and quantification of polyphenols from chloroform and methanol fractions of Cucumis melo. ICP-OES was used for quantitative and qualitative evaluation of elements, and results showed that magnesium (533.6-663.2 ppm) and iron (308.1-590.8 ppm) were the most abundant minerals. Amino acids analyzer showed that valine (0.03-45.58 mmol/L) and taurine (0.03-2.34 mmol/L) were present in the highest concentration. The methanol extract yield of the accessions was in the range of 14.00-17.03 g/100 g of plant material. HPLC analysis showed the detection of eleven phenolic compounds, and the most abundant compound was chlorogenic acid (1120.53–1237.52  $\mu$ g/g), followed by gallic acid  $(188.14-930.74 \ \mu g/g)$  and vanillic acid  $(6.35-298.81 \ \mu g/g)$ . Chloroform and methanol fractions were analyzed for antioxidant activity, and results showed that all the fractions were rich in total phenolic content (TPC) (26.6-112.00 mg/100 g) and total flavonoid content (TFC) (20.5-77.9 mg/100 g). Methanol fraction of accession-1 showed the best free radical scavenging capacity (82.2 %) and % inhibition of linoleic acid peroxidation activity (84.4 %) than other fractions. Statistical analysis revealed significant variation among the nutritional potential and antioxidant activity of all the accessions.

## 1. Introduction

Food, as well as nutritional security, has also lately been identified as one of the world's most persistent problems. Especially in developing nations, low food consumption and limited availability of food remain unsolved concerns (Ercisli and Sagbas, 2017). Moreover, the rapid expansion in the world's population (expected to be 9.6 billion in 2050) caused significant increases in food prices and boosted the land degradation (Meena et al., 2022) that further complicated the food insecurity conditions. So, in view of the present estimates the use of conventional food crops will not be enough to fulfill the world food requirements. Therefore, there is an intense need to explore other non-conventional food sources to accomplish the world food demand and to reduce pressure on conventional food crops (Falade et al., 2020).

The study of underutilized wild plants produced in natural populations can bring new insights that could lead to the introduction of new populations with better composition of secondary metabolites, which are advantageous for various food and pharmaceutical applications (Fallah et al., 2020, Scarano et al., 2021). Furthermore, utilization of mineral, amino acids, and polyphenols-rich food, also called functional food, can provide strength to combat oxidative stress (Iftikhar et al., 2022). Thus, there is a significant scientific interest in the study of underutilized wild plants that are rich in polyphenols to fulfil food insecurity requirements and might be employed in the food and pharmaceutical industries (Stojiljković et al., 2016).

Cucumis melo var. Agrestis, often known as wild melon, is an

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underutilized vine plant that may grow up to 1.5 m in height. It is an annual or perennial plant and belongs to the Cucurbitaceous family (Hui et al., 2020; Memon et al., 2018). Cucumis melo var. Agrestis exhibits a versatile capacity to grow under various environmental conditions. The cultivation occurs under a consistent temperature range of 25-40 °C (Tanveer et al., 2012). Cucumis melo var. Agrestis fruit are rich in tannins, flavonoids, alkaloids, phenolic components, steroids, resins, terpenoids, glycosides, and saponins (Arunachalam and Chinnaraju, 1970, Memon et al., 2018, Kapoor et al., 2020; Gopalasatheeskumar and Kalaichelvan, 2021). Fruit is also source of important minerals like sodium, potassium, zinc, copper, and iron, as well as some important amino acids (Gopalakrishnan and Thangaraj, 2014). The traditional medicinal uses of Cucumis melo var. Agrestis have inspired many pharmacological studies including anti-inflammatory, diuretic, antioxidant (Naik et al., 1980, Salahuddin and Jalalpure, 2010, Arunachalam and Chinnaraju, 2012, Gopalasatheeskumar et al., 2019, Pratima, 2019). Although few reports on the nutritional potential of Cucumis melo var. Agrestis are available in literature but to the best of our knowledge, no data regarding the variations in phenolic profile, nutritional composition, and biological activities among different accessions of the plant, native to Pakistan is reported. Therefore, the aim of the present study was to investigate the phenolic profile, mineral contents, and amino acid composition along with the antioxidant potential of various accessions of wildly grown Cucumis melo var. Agrestis. Native to Pakistan.

# 2. Materials and methods

All the chemicals and reagents used in this study were of analytical quality and purchased from Sigma-Aldrich Chemical Co. (St. Louis, MO) and Merck (Darmstadt, Germany), respectively, unless otherwise stated.

# 2.1. Collection, identification, and pretreatment of plant materials

Various accessions of wildly grown *Cucumis melo* var. *Agrestis* (fully ripened) were collected from Layyah Desert (30–45 to 31–24 degrees North latitudes and 70–44 to 71–50 degrees East longitudes), South Punjab, Pakistan (Fig. 1). The species were identified and authenticated by a Taxonomist, Department of Botany, Government College University Faisalabad, Pakistan. The fruits were shade-dried for almost a week, ground into fine powder (80-mesh) with a grinder (TSK-949, Westpoint, France), and stored in sealed polythene containers at room temperature for subsequent study.

# 2.2. Nutritional potential

# 2.2.1. Mineral composition

The crude powdered sample was digested according to the procedure described by (Wolf, 1982) with slight modifications. Briefly, 0.2 g of powdered material in 2 mL of digestion mixture was left overnight. The solution was subjected to heating at 110–200 °C until the vapors generated at that temperature had completely evaporated. Perchloric acid (HClO<sub>4</sub>) was added until the color of the liquid changed from black to dark brown and then to colorless. The digestion flasks were then allowed to cool, contents were filtered and at the end make the volume up to 50 ml. Then, the samples were analyzed by ICP-OES (Prodigy, Teledyne, USA). Metal solutions (0.5–1000 mg/mL) were prepared, and calibration curves of each metal were drawn.

# 2.2.2. Amino acid analysis

Amino acid analysis was performed on an amino acid analyzer following the reported method by Iriti et al. (2009). Briefly, 1 g of sample was added to 5 mL of chilled 80 % ethanol and centrifuged (14,000 rpm) for 10 min at 4 °C. Separate the supernatant and mix with 1 mL of 3 % sulfosalicylic acid (SSA) and vortexed. Kept the mixture overnight and centrifuged it under the same conditions as mentioned above. Dried the sample with nitrogen reflux and then added a buffer in the known quantity of the sample. The sample was analyzed on a Biochrom 30 + amino acid analyzer using a lithium-ion exchange column and ninhydrin post-derivatization method. The quantification was done using the standard addition method.

# 2.3. Preparation of extracts and fractionation

The extraction of the shade-dried and pulverized material (20 g) was subjected to a 500 mL Soxhlet unit for 4 h. A rotary evaporator was used to remove the solvents from sample extracts under low pressure (Hussain et al., 2013). The dried plant extracts were subjected to fractionation sequentially using three different polarity solvents: n-hexane, chloroform, and methanol, using the flow sheet diagram given below (Fig. 2). The yield of the fractions was calculated, and fractions were stored at -4 °C in glass vials for further analysis.

# 2.4. RP-HPLC analysis of phenolic compounds

The analysis of polyphenols from plant extracts was conducted using



Fig. 1. Fruits of various accessions of Cucumis melo var. Agrestis.



Fig. 2. Protocol for the fractionation of extracts of various accessions of Cucumis melo var. Agretis.

HPLC with a C18 column, as reported previously by Hussain et al. (2013). The column had dimensions of  $250 \times 4.6$  mm and a particle size of 5 µm. The Flexer Chromera HPLC system from Perkin Elmer, USA, was utilized. It consisted of a binary LC pump and a UV–visible LC Detector from Shelton CT, 06484 USA. System was managed by the software V. 4.2. 6410. The experiment employed a gradient system consisting of solvent A, which was a mixture of acetonitrile and methanol at a ratio of 70:30, and solvent B, which was double distilled water containing 0.5 % glacial acetic acid. Column over temperature was set at 30 °C and UV spectra were taken at 275 nm. The analytes were detected by spiking samples with standards and matching retention times, and further their quantifications were done by using an external standard method.

# 2.5. Evaluation of antioxidant activity

Antioxidant activities of chloroform and methanol fractions (1 mg/ ml) of various accessions of Cucumis melo var. Agrestis were evaluated through various antioxidant activities. The quantity of total phenolic content (TPC) of fractions (10 mg/mL) were figured out using FC reagent as reported by Hussain et al. (2012). The absorbance values at different concentrations of gallic acid were used to construct the calibration curve, and the regression equation (y = 2.7974x + 0.5155; R2 = 0.9826) from the calibration curve was used to figure out the TPC from the extracts. Total flavonoid content (TFC) of chloroform and methanol fractions (20 µg/ml) was studied by using the method reported by Hussain et al. (2012). The calibration curve regression equation, y =0.7242x - 0.0466; R2 = 0.998, was used to estimate the amount of total flavonoid content in the extracts. This was shown as milligrams of quercetin equivalents (QE) per 100 g weight of dry fruit (mg/100 g). To measure the DPPH radical scavenging activity of extracts (50 µg/mL), the protocol given by Hussain et al. (2011), was followed, and BHA was employed as positive control. Bleachability of  $\beta$ -carotene in the linoleic acid system was evaluated to measure the inhibition using the protocol reported by Hussain et al. (2011). Evaluation of reducing power assay of the fractions was done using a method reported by Hussain et al. (2013).

# 2.6. Statistical analysis

Three samples were taken from each accession and individually evaluated in a triplicate manner. The information was presented as mean values  $\pm$  standard deviation. The STATISTICA 5.5 software was used to perform the Analysis of Variance (ANOVA) test. The difference was considered statistically significant if the probability (p) value is  $\leq$  0.05.

# 3. Result and discussion

# 3.1. Nutritional potential

# 3.1.1. Mineral composition

Comparative mineral composition of the various accessions of *Cucumis melo* var. *Agrestis* analyzed in this study are shown in Table 1. ICP-OES analysis revealed that Mg, Fe, Cu, and Cd were the major minerals while Si, Co, Ni, Pb, Cr, Mn, and Zn were the minor in the

Table	1

Mineral composition of various accessions of Cucumis melo var. Agrestis.

Minerals	Concentration (ppm)							
	Accession-1	Accession-2	Accession-3	Accession-4				
Mg	$663.2 \pm 18.0^{\mathrm{b}}$	$587.2\pm20.0^{\text{a}}$	$533.6\pm16.2^{\text{a}}$	$653.1\pm16.3^{b}$				
Fe	$590.8 \pm 13.0^{\rm c}$	$415.0\pm18.6^{\rm b}$	$308.1\pm12.0^{\rm a}$	$432.2\pm20.9^{\rm b}$				
Cu	$39.10 \pm 1.06^{\mathrm{a}}$	$40.16\pm1.02^{a}$	$39.44 \pm 1.84^{\mathrm{a}}$	$39.23\pm0.98^{\rm a}$				
Cd	$14.21\pm1.01^{\rm b}$	$12.34\pm1.07^{\rm ab}$	$11.27\pm0.93^{\rm b}$	$13.13\pm0.85^{\rm ab}$				
Si	$3.51\pm0.23^{a}$	$3.63\pm0.20^{a}$	$\textbf{4.47} \pm \textbf{0.31}^{\rm b}$	$3.55\pm0.25^a$				
Со	$1.49\pm0.91^{a}$	$1.70\pm0.15^{\rm a}$	$1.93\pm0.10^{\rm a}$	$1.71\pm0.13^{\rm a}$				
Ni	$0.68\pm0.04^a$	$0.69\pm0.03^a$	$0.66\pm0.03^a$	$0.72\pm0.05^{a}$				
Pb	$0.92\pm0.24^{\rm b}$	$0.90\pm0.03^{\rm b}$	$0.75\pm0.05^a$	$0.75\pm0.03^{a}$				
Cr	< 0.12	< 0.12	< 0.12	$0.21\pm0.02^{a}$				
Mn	< 0.12	< 0.12	< 0.12	< 0.12				
Zn	< 0.12	< 0.12	< 0.12	< 0.12				

The values are reported as mean  $\pm$  SD of three independent experiments. Different superscript letters show significant (p  $\leq$  0.05) difference among accessions.

Cucumis melo var. Agrestis (Table 1). Mg and Fe were the most abundant and major minerals detected from all the accessions. Mg and Fe showed concentration ranges from 533.6-663.2 ppm and 308.1-590.8 ppm, respectively. Co and Cd were also found in major amounts from 39.10-40.16 ppm and 11.27-14.21 ppm, respectively. Other elements, i. e., Si, Co, Ni, and Pb, were present in minor amounts in the accessions. Zn, Cr, and Mn were found in traces and below the limit of detection (<0.12 ppm). Overall, accession-1 showed a significant concentration of essential minerals. Statistical analysis showed significant (p  $\leq$  0.05) variations in the concentration of elements among various accessions of Cucumis melo. According to numerous studies, magnesium intake from the daily diet is insufficient, and magnesium plays a vital role in our body, especially at the cellular level. While iron is involved in many metabolic processes, and iron deficiency is a serious issue for adults, particularly women (Abbaspour et al., 2014, Fiorentini et al., 2021). So, Cucumis melo var. Agrestis can be the cheapest source of magnesium and iron. In the literature, few reports are available on the mineral content of Cucumis melo var. Agrestis. Fruit contains (mg per total ash): Mg, 15.4; Cu, 1.03; Ca, 19.6; Na, 308; Co, 0.47; P, 10; Fe, 0.28; and Ni, 0.37 (Dahot et al., 1999, Memon et al., 2018). In another report, the major mineral metals component in Cucumis melo var. Agrestis fruit analyzed was Zinc (17.6 ppm), Ferrous (141 ppm), Iron (14.3 ppm), and Copper (13.4 ppm) (Memon et al., 2018). Hence, our results are consistent with prior findings. Various accessions of Cucumis melo var. Agrestis have mineral concentrations even higher than values reported from some of the important Cucurbitaceae plants. The iron concentration is higher than Lagenaria siceraria (118.7 ppm), Citrullus lanatus (211.3 ppm), and Citrullus colocynthis (191.3 ppm). Magnesium concentration is also higher than Citrullus lanatus (123.9 ppm) and Citrullus colocynthis (79.75 ppm), but lower than Lagenaria siceraria (1623.3 ppm). The concentration of copper is also higher than Lagenaria siceraria (1.90 ppm), Citrullus lanatus (31.6 ppm) and, Citrullus colocynthis (20.12 ppm) (Shah et al., 2010, Falade et al., 2020).

#### 3.1.2. Amino acid analysis

The amino acid compositions of the accessions of Cucumis melo var. Agrestis were expressed as mmol/L and are presented in Table 2. In total, 17 amino acids were identified, including essential and non-essential ones. The highest values among amino acids were valine (0.03-45.58 mmol/L) and taurine (0.03-2.34 mmol/L) followed by isoleucine (0.03-2.26 mmol/L) and leucine (0.01-2.27 mmol/L). Citrulline (0.01-0.29 mmol) and 4-aminobutyric acid (0.01-0.19 mmol/L) were found in minor concentrations. Among all accessions, sarcosine,

#### Table 2

threonine, cysteine, and tyrosine amino acids were also identified in small concentrations. Statistical analysis showed significant variation (p  $\leq$  0.05) in the amino acid composition of all the accessions of *Cucumis* melo var. Agrestis. Table 3

Amino acid analyzer allows far faster and more dedicated analysis of amino acids than prior technologies that used UV or fluorescence detection (Shafaei et al., 2011). Both essential and non-essential amino acids play a crucial function in the human body such as in endothelial cells and non-vascular tissues e.g. arginine, glutamine, alanine, taurine, and glycine enhance CO production through heme oxygenase, also arginine, glutamate, and proline participate in many of the cell signaling pathways (Takahashi et al., 2011). Hence, Cucumis melo var. Agrestis can be the cheapest source of many of the essential and non-essential amino acids. There is limited literature available that reports the identification and quantification of amino acids in Cucumis melo var. Agrestis. To the best of our knowledge, no data has been published that reports the free amino acid content in various accessions of Cucumis melo var. Agrestis.

# 3.2. Yield of fruit extract

The extracts vield from the various accessions of *Cucumis melo* fruits was 14.00-17.03 g/100 g of dry plant material (Table 1). Yields of hexane, chloroform, and methanol fractions were in the range of 0.96-2.35, 3.15-4.53, and 9.09-11.45 g/100 g of dry plant material, respectively. Overall, the yield of extracts from different fractions increased in the manner of low-polarity to high-polarity solvents. The

#### Table 3

Yield of extract and different fractions from various accessions of Cucumis melo var. Agrestis.

Fruit	Extract Yield	Fractions (g/100 g)				
Accession	(g/100 g)	Hexane	Chloroform	Methanol		
Accession-1	$\begin{array}{c} 17.03 \pm \\ 1.10^{\mathrm{b}} \end{array}$	$\begin{array}{c} 0.96 \pm \\ 0.06^{\mathrm{b}} \end{array}$	$4.53\pm0.12^{c}$	$11.45\pm0.69^{\text{c}}$		
Accession-2	$14.00\pm1.11^a$	$\begin{array}{c} 0.83 \pm \\ 0.10^{a} \end{array}$	$3.15\pm0.13^{a}$	$\begin{array}{c} 10.04 \pm \\ 0.76^{bc} \end{array}$		
Accession-3	$\begin{array}{c} 15.40 \ \pm \\ 0.81^{ab} \end{array}$	$\begin{array}{c} 1.84 \pm \\ 0.12^c \end{array}$	$3.92\pm0.13^{b}$	$9.65\pm0.28^{b}$		
Accession-4	$14.70\pm0.84^a$	$2.35 \pm 0.10^{d}$	$\textbf{3.24}\pm0.09^{a}$	$9.09\pm0.19^a$		

The results represent the average of three independent experiments with standard deviations. Different superscript alphabet letters show significant (p  $\leq$ 0.05) difference among accessions.

Amino acids	Retention time (min)	Concentration (mmol/L)					
		Accession-1	Accession-2	Accession-3	Accession-4		
Taurine	4.43	$2.34\pm0.13^{\rm d}$	$1.85\pm0.11^{\rm c}$	$0.60\pm0.05^{b}$	$0.03\pm0.01^a$		
Aspartic acid	16.23	$1.29\pm0.08^{\rm b}$	_	$0.06\pm0.01^{\rm a}$	-		
Threonine	19.50	$0.13\pm0.01^{\rm c}$	$0.11\pm0.01^{\rm bc}$	$0.02\pm0.00^{\rm a}$	$0.09\pm0.01^{\rm b}$		
Serine	20.83	$0.55\pm0.03^{\rm c}$	$0.03\pm0.01^{a}$	$0.04\pm0.01^{a}$	$0.22\pm0.02^{\rm b}$		
Asparagine	24.36	$0.08\pm0.01^{\rm a}$	$0.14\pm0.01^{\rm b}$	$0.16\pm0.01^{\rm b}$	$1.41\pm0.08^{\rm c}$		
Glutamic acid	26.60	$0.29\pm0.01^{\rm d}$	$0.20\pm0.10^{\rm c}$	$0.15\pm0.01^{\rm b}$	$0.01\pm0.00^{\rm a}$		
Sarcosine	31.76	$0.01\pm0.00^{\rm a}$	$0.02\pm0.01^{\rm a}$	$0.08\pm0.01^{\rm b}$	$0.27\pm0.01^{\rm c}$		
Citruline	39.40	_	$0.29\pm0.01^{\rm c}$	$0.01\pm0.00^{\rm a}$	$0.07\pm0.01^{\rm b}$		
Valine	42.13	$8.54 \pm 1.02^{\rm b}$	$10.68\pm1.06^{\rm b}$	$0.03\pm0.01^{\rm a}$	$45.58 \pm \mathbf{1.27^c}$		
Cysteine	51.00	$0.01\pm0.00^{\rm a}$	$0.01\pm0.00^a$	_	$0.01\pm0.00^{\rm a}$		
Methionine	51.23	_	_	_	$\textbf{0.04} \pm \textbf{0.01}$		
Isoleucine	54.60	$1.86\pm0.07^{\rm c}$	$2.26\pm0.09^{\rm d}$	$0.03\pm0.01^{\rm a}$	$0.20\pm0.02^{\rm b}$		
Leucine	55.76	$1.85\pm0.08^{\rm c}$	$2.27\pm0.11^{\rm d}$	$0.01\pm0.01^{\rm a}$	$0.20\pm0.01^{\rm b}$		
Tyrosine	59.43	$0.11\pm0.01^{\rm b}$	$0.02\pm0.01^{\rm a}$	$0.01\pm0.01^{\rm a}$	$0.02\pm0.01^{\rm a}$		
β-Alanine	60.46	$1.75\pm0.07^{\rm b}$	$3.68\pm0.21^{\rm c}$	_	$0.30\pm0.02^{\rm a}$		
4-Aminobutyric acid	68.33	_	$0.19\pm0.01^{\rm c}$	$0.01\pm0.00^{\rm a}$	$0.04\pm0.01^{b}$		
Lysine	81.20	$0.02\pm0.01^a$	_	-	$0.05\pm0.01^{b}$		

The values are reported as mean  $\pm$  SD of three independent experiments.

Different letters in superscript represented significant different among the different accessions.

*Cucumis melo* fruit extracts exhibited significant concentrations of polar compounds, including flavonoids and phenolics. Gopalasatheeskumar and Kalaichelvan, (2021), reported the percentage yield of *Cucumis melo* var. *Agrestis* was 18.22 % with the cold maceration extraction method. There is currently no literature that provides information on the yield of distinct solvent-based fractions from various accessions of *Cucumis melo* var. *Agrestis*. A minimal difference in extract yield could be attributed to the availability of various extractable components using different solvents and extraction techniques (Hussain et al., 2012, Iqbal et al., 2013).

# 3.3. Phenolic acid and flavonoid profile

Polar fractions of accessions of *Cucumis melo* var. *Agrestis* were qualitatively and quantitatively analyzed for phenolic acid and flavonoids using the RP-HPLC and results are presented in Table 4. Phenolic acids include caffeic acid, chlorogenic acid, ferulic acid, gallic acid, hydroxybenzoic acid, kaempferol, *p*-coumaric acid, quercetin, sinapic acid, salicylic acid, vanillic acid, While Gallic acid (188.14–930.74 µg/g) and vanillic acid (6.35–298.81 µg/g) were the major phenolics in all the accessions. *p*-coumaric acid (33.28 µg/g), kaempferol (3.86–563.24 ppm) and quercetin (9.77–239.02 µg/g) were present in moderate concentrations. Small amounts of caffeic acid and sinapic acid were detected. In comparison, approximately all the accession 1 has a comparatively smaller number of phenolics. Sinapic acid was only found in accession 3 and 4. HPLC chromatograms of accession 3 and 4 chloroform, and methanol fractions are shown in Fig. 3.

Dietary phenolics are garnering a lot of attention now because of their potential antioxidative and anticarcinogenic effects. Free radical scavengers, reducing agents, and scavengers of singlet oxygen production are other roles played by phenolic compounds (Ghasemzadeh and Ghasemzadeh, 2011). Hence, many phenolic compounds in *Cucumis melo* var. *Agrestis* depicts the nutritive value of this cheap fruit. A literature study showed that *Cucumis melo* var. *Agrestis* has different phenolic acids in its fruit and seeds. Vanillic acid is the major phenolic compound 2.3–2.9 % in it, followed by callistephin and catechin (Mariod and Matthaeus, 2008). Furthermore, there is a lack of data on the detection and measurement of phenolic compounds in the fractions of various *Cucumis melo* var. *Agrestis* accessions.

#### 3.4. Antioxidant activity

# 3.4.1. Total phenolic and total flavonoid contents

The total phenolic contents of the extracts were estimated using a calibration curve and represented as mg of gallic acid equivalents (GAE) per 100 g of dry weight of fruit (mg/g), as shown in Table 5. TPC of chloroform (CH<sub>3</sub>Cl) and methanol fractions (CH<sub>3</sub>OH) ranges from

26.6–112.0 mg/100 g of dry fruit material measured as GAE. Accession 1 showed highest while accession 2 showed the lowest TPC. Total flavonoid contents (TFC) in chloroform and methanol fractions of various accessions of *Cucumis melo* var. *Agrestis* were determined and reported in Table 5 as mg/100 g of dry material, measured as QE. TFC of chloroform (CH<sub>3</sub>Cl) and methanol fractions (CH<sub>3</sub>OH) ranges from 20.5–77.9 mg/100 g of dry fruit materials, measured as QE. Total flavonoid contents were highest in accession 1, while lowest in accession 3. Our results showed variation in the TPC and TFC of different fractions of *Cucumis melo* var. *Agrestis* accessions were significant ( $p \le 0.05$ ).

Phenolics are present in nearly all plants, and their antioxidant property has important physiological and morphological benefits. As a result, quantifying phenolic content and assessing its contribution to oxidative stress is important (Iftikhar et al., 2022). Limited literature exists regarding the TPC and TFC of *Cucumis melo* var. *Agrestis* fruit; however, no report has yet been compiled that conducts a comparative analysis of the fractions from different accessions (Arora et al., 2011, Gill et al., 2011, Sahithi et al., 2015, Memon et al., 2018). According to the literature, the fruit of *Cucumis melo* var. *Agrestis* contained 19.82 mg/g TPC, calculated as GAE, and 18.39 mg/g TFC, calculated as QE (Gopalasatheeskumar and Kalaichelvan, 2021).

# 3.4.2. DPPH radical scavenging assay and inhibition of linoleic acid peroxidation

The DPPH radical scavenging antioxidant assay was used to assess the antioxidant activity of chloroform and methanol fractions (50 µg/ mL) of various accessions of *Cucumis melo* var. *Agrestis in vitro*, while the results are reported as % radical scavenging activity (% RSA) in Table 5. All the extracts were able to quench 58.5–82.2 % DPPH free radicals. Accession 1 showed the best radical scavenging capacity (82.2 %). Methanol (CH<sub>3</sub>OH) fraction generally showed significantly higher RSA than Chloroform (CH<sub>3</sub>Cl) fractions. Antioxidant properties are basically influenced by phenolic components, and the highest activity of accession 1 methanol extract may be correlated with the highest TPC and TFC of the extracts. Due to an increase in the concentration of phenolic components, the DPPH free radical scavenging ability of extracts rises as its concentration rises (Iftikhar et al., 2022).

Literature review showed that there is no data available on DPPH free radical scavenging activity for fractions of various accessions of *Cucumis melo* var. *Agrestis*. According to Arora et al., 2011, seed extract had a strong DPPH scavenging action of 75.59 % at 300  $\mu$ g/mL and 69.86 % at 400  $\mu$ g/mL. The percent inhibition of linoleic acid peroxidation by different fractions of *Cucumis melo* var. *Agrestis* accessions (10 mg/mL) are shown in Table 5. Accessions 1 and 3 showed the highest while accession 2 showed the lowest percent inhibition. Percent inhibition in linoleic acid peroxidation by various accessions fractions of *Cucumis melo* var. *Agrestis* ranged from 65.01–84.35 %. Methanol

#### Table 4

Quantification of phenolic acid and flavonoids from polar fractions of various accessions of Cucumis melo var. Agrestis.

Compounds	Concentration (µg/g)							
	Accession-1		Accession-2		Accession-3		Accession-4	
	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH
Chlorogenic acid	_	-	_	1237.52 <sup>a</sup>	_	1120.53 <sup>a</sup>	_	_
p-cumaric acid	_	511.84 <sup>d</sup>	$88.12^{b}$	_	_	_	138.10 <sup>c</sup>	33.28 <sup>a</sup>
Gallic Acid	308.33 <sup>c</sup>	188.14 <sup>a</sup>	384.33 <sup>d</sup>	217.31 <sup>b</sup>	$654.2^{f}$	328.45 <sup>c</sup>	422.74 <sup>e</sup>	930.74 <sup>g</sup>
Hydroxy benzoic acid	_	-	-	33.45	-	_	-	-
Caffeic Acid	7.76 <sup>b</sup>	-	-	_	3.61 <sup>a</sup>	_	3.61 <sup>a</sup>	_
Vanillic acid	27.64 <sup>c</sup>	6.35 <sup>a</sup>	68.25 <sup>d</sup>	_	_	$20.49^{b}$	298.81 <sup>e</sup>	
Kaempferol	_	-	39.96 <sup>c</sup>	$12.36^{b}$	-	$13.40^{b}$	563.24 <sup>d</sup>	3.86 <sup>a</sup>
Ferulic acid	$12.75^{a}$	-	-	_	$43.72^{b}$	_	_	56.86 <sup>c</sup>
Sinapic acid	_	-	-	_	9.44 <sup>b</sup>	_	_	1.93 <sup>a</sup>
Salicylic acid	1.89 <sup>a</sup>	-	257.88 <sup>c</sup>	_	13.27 <sup>b</sup>	_	_	_
Quercetin	9.77 <sup>a</sup>	-	_	167.32 <sup>c</sup>	_	_	27.73 <sup>b</sup>	239.02 <sup>d</sup>

The results represent the average of three independent experiments. The SD was <10 % in all the values.

Different superscript letters show significant (p  $\leq$  0.05) difference among fractions.



Fig. 3. Typical Chromatogram showing the separation of phenolic acids and flavonoids from chloroform fraction of accession 4 of Cucumis melo.

# Table 5

Antioxidant activity and radical scavenging capacity of chloroform and methanol fractions of Cucumis melo var. Agrestis accessions.

Antioxidant Assays	Accession-1		Accession-2		Accession-3		Accession-4		Standard
	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	CH <sub>3</sub> Cl	CH <sub>3</sub> OH	BHA
TPC (mg/100 g) <sup>α</sup>	$80.3\pm2.1^{d}$	$\begin{array}{c} 112.0 \ \pm \\ \textbf{4.8}^{\mathrm{f}} \end{array}$	$26.6 \pm 1.1^{a}$	$68.2 \pm \mathbf{1.9^{c}}$	$\begin{array}{c} \textbf{72.2} \pm \\ \textbf{2.8}^{c} \end{array}$	$98.4\pm3.1^{\text{e}}$	$\begin{array}{c} 62.3 \pm \\ 2.2^{\mathrm{b}} \end{array}$	$\begin{array}{c} 105.3 \pm \\ 2.5^{\mathrm{f}} \end{array}$	_
TFC $(mg/100 g)^{\beta}$	$\textbf{47.4} \pm \textbf{1.4}^{d}$	$\textbf{77.9} \pm \textbf{1.6}^{f}$	$\begin{array}{c} 34.2 \pm \\ 1.0^{b} \end{array}$	$52.2\pm1.6^{e}$	$\begin{array}{c} 20.5 \pm \\ 1.0^{a} \end{array}$	$\textbf{48.5} \pm \textbf{1.5}^{d}$	$\begin{array}{c} \textbf{38.2} \pm \\ \textbf{1.0}^{c} \end{array}$	$\textbf{75.2} \pm \textbf{2.1}^{\rm f}$	_
DPPH (% RSA) <sup>γ</sup>	$\begin{array}{c} \textbf{75.2} \pm \\ \textbf{2.2}^{bc} \end{array}$	$82.2 \pm 2.6^{d}$	$\begin{array}{c} 62.3 \pm \\ 2.6^{\rm a} \end{array}$	$\textbf{74.3} \pm \textbf{2.8}^{b}$	$\begin{array}{l} 58.5 \pm \\ 1.9^{\rm a} \end{array}$	$\begin{array}{c} \textbf{76.4} \pm \\ \textbf{2.8}^{bc} \end{array}$	$\begin{array}{c} \textbf{70.2} \pm \\ \textbf{2.7}^{b} \end{array}$	$78.2 \pm \mathbf{1.1^c}$	$\textbf{97.1} \pm \textbf{2.5}^{e}$
Inhibition of linoleic acid peroxidation $(\%)^{\delta}$	$\begin{array}{c} \textbf{77.2} \pm \\ \textbf{2.0}^{\rm bc} \end{array}$	$84.4\pm3.0^{c}$	${}^{65.0\pm}_{2.1^a}$	$\begin{array}{c} \textbf{71.2} \pm \\ \textbf{2.9}^{ab} \end{array}$	$\begin{array}{c} 68.4 \pm \\ 1.9^{\mathrm{a}} \end{array}$	$\begin{array}{c} 80.2 \pm \\ 2.2^{bc} \end{array}$	$\begin{array}{c} \textbf{70.1} \pm \\ \textbf{1.8}^{a} \end{array}$	$\textbf{75.2} \pm \textbf{2.6}^{b}$	$\begin{array}{c} 94.3 \pm \\ 2.0^d \end{array}$

The values are reported as mean  $\pm$  SD of three independent experiments.

Different superscript alphabet letters show significant (p  $\leq$  0.05) difference among fractions.

 $^{\alpha}\,$  Total phenolic contents, mg/100 g of dry fruit material measured as GAE.

 $^\beta$  Total flavonoid contents mg/100 g of dry fruit materials, measured as QE.

 $^{\gamma}\,$  % radical scavenging by 10 mg/mL extract solution.

 $^{\delta}\,$  % inhibition of linoleic acid peroxidation by 10 mg/mL extract solution.

(CH<sub>3</sub>OH) fractions of all the accessions showed significantly higher percent inhibition of linoleic acid peroxidation than Chloroform (CH<sub>3</sub>Cl) fractions. The result showed that fractions rich in TPC and TFC inhibited linoleic acid peroxidation to a greater extent. No earlier reports are available regarding the data availability on percent inhibition in the linoleic acid system assay for *Cucumis melo* var. *Agrestis* fractions.

# 3.4.3. Evaluation of reducing power assay

The trends of reducing potentials of different factions of *Cucumis melo* var. *Agrestis* accessions are presented in Fig. 4. The figure illustrates a positive correlation between absorbance and the antioxidant activity and reduction potential of the samples. At concentrations of up to 10 mg/ml, the reducing power of various fractions of *Cucumis melo* var. *Agrestis* accessions are evaluated. The reducing potential of samples was proportional to their extract concentration. The reducing power of different fractions of extracts lowered in the order: Standard (Ascorbic

acid) > accession 1 methanol > accession 1 chloroform > accession 3 chloroform > accession 2 methanol > accession 2 chloroform > accession 4 methanol > accession 3 methanol > accession 4 chloroform. When compared with ascorbic acid (standard), all the fractions showed significantly ( $p \leq 0.05$ ) less reducing power.

Limited data are available on the ferric reducing power of *Cucumis melo* var. *Agrestis*. In a previous study, it was reported that the ferric reducing power of fruit extract was  $30.77 \,\mu$ g/mL, while for seeds,  $28.13 \,\mu$ g/mL was calculated in terms of ascorbic acid equivalent (Gopalasa-theeskumar and Kalaichelvan, 2021). However, there is no report available for ferric reducing power on different fractions of various accessions of *Cucumis melo* var. *Agrestis*.

## 3.5. Conclusion

The present study highlighted the untapped potential of Cucumis



Fig. 4. Evaluation of reducing power of different fractions of Cucumis melo var. Agrestis accessions.

melo var. Agrestis, a wild fruit rich in minerals, amino acids, and phenolic compounds. It can be concluded that all the accessions were rich in important minerals and amino acids. Magnesium and iron were the most abundant minerals found in all the accessions. Other minerals detected in high concentration were copper, cadmium, silicon, and cobalt. A range of both essential and non-essential amino acids were detected from all the accessions. The major amino acid was valine and accession 4 has maximum concentration of valine followed by accession 2 and 1. The amino acids found in Cucumis melo var. Agrestis are essential for protein synthesis, metabolic activities, and overall good health, highlighting the promising potential of this fruit as a functional food ingredient. Moreover, HPLC study revealed presence of 11 phenolic compounds. Qualitative and quantitative variations in the phenolic acid and flavonoid contents of all the accessions were recorded. Chlorogenic acid was the most abundant phenolic acid identified in accession 2 and 3. Gallic acid was detected in all the accessions. Kaempferol and quercetin were highest in accession 4 followed by accession 2 of Cucumis melo. All the accessions exhibited antioxidant activity in terms of TPC, TFC, DPPH radical scavenging, inhibition of linoleic acid peroxidation and reducing potential. Methanol fraction of accession 1 showed the highest TPC and TFC and thus best DPPH radical scavenging and percent inhibition of linoleic acid peroxidation activities. extracts, as compared to the chloroform fractions. These results provide good evidence to explore Cucumis melo var. Agrestis, as potent agents for the antioxidant and nutritional properties and can be utilized for food products. Further research can be conducted to investigate the antioxidant potential of Cucumis melo extracts in the various food systems.

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#### CRediT authorship contribution statement

Hira Zulfiqar: Writing - original draft, Methodology, Investigation,

Formal analysis. Abdullah Ijaz Hussain: Writing – review & editing, Supervision, Project administration, Methodology, Conceptualization. Qasim Ali: Writing – original draft, Software, Conceptualization. Hassaan A. Rathore: Writing – review & editing, Supervision, Project administration, Methodology, Conceptualization. Iqbal Ahmed: Writing – original draft, Supervision, Software, Methodology.

#### **Declaration of Competing Interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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