### SYNTHESIS AND BIOLOGICAL ACTIVITY OF SOME NEW 4-(AMINOACYL) AMINOPYRIDINES AND 2-(AMINOACYL) AMINOPYRIMIDINE DERIVATIVES

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Key Words: 4 - (Aminoacyl) aminopyridines and 2 - (Aminoacyl) aminopyrimidine Derivatives.

### **ABSTRACT**

The synthesis of 4-(N-Tos and N-Pht-aminoacyl) aminopyridines (III - XIV) and 2-(N-Pht-aminoacyl)-aminopyrimidines (XV-XXIV) has been achieved employing the acid chloride and carbodiimide methods. Hydrazinolysis of 4-(N-Pht-Gly or -B-Ala-) aminopyridines or 2-(N-Pht-L-Phe- or -B-Ala-) aminopyrimidines in ethanol afforded the desired 4-(Gly- or B-Ala) aminopyridines (XXV - XXVI) and 2-(L-Phe- or B-Ala) aminopyrimidines (XXVIII - XXVIII) respectively. 4-(N-Pht- or N-Tos-dipeptidyl) aminopyridines (XXIX - XXXVI) are synthesized via the DCC method and 2-(N-Tos-L-Val-L-Leu) aminopyrimidine (XXXVII) via the azide method. The amino acid derivatives (XV, XVI, XXI, XXII, XXV, XXVI, XXVIII) and the dipeptide (XXXVII) are found to be active against a number of microorganisms.

### INTRODUCTION

The interesting pharmacological properties of 2- aminopyrimidine and 4- aminopyridine derivatives (Abramovitch, 1974 and Brown, 1970) suggested the possibility of potential activity of simple 2-(aminoacyl) aminopyrimidine and 4-(aminoacyl) aminopyridine derivatives. It seem therefore desirable to synthesize some 4-(aminoacyl) aminopyridine and 2-(aminoacyl)-aminopyrimidine derivatives which may be of verified or intensified pharmaceutical effects. In continuation of our previous work (El-Naggar and Zaher 1977, El-Naggar and Ismail 1977 and El-Naggar et al 1981) the synthesis and microbiological studies of some 4-(N-Tos-or N-Pht-aminoacyl or aminoacyl or N-Pht- or N-Tos-dipeptidyl) amino pyridines and 2-(N-Tos-or N-Pht-aminoacyl or aminopyrimidines or N-Tos-L-Tos-L-Val-L-Leu) aminopyridines are reported in this paper.

### **EXPERIMENTAL**

Melting points reported are uncorrected. TLC was carried out using silica gel-G and developed with benzene-ethyl acetate (1:1) mixture. Visualization of spots was done by spraying with iodine-potassium iodide (20%) solution. Benzidine, ninhydrin, silver nitrate and hydroxamate reactions were used as visualizing reagents (PC-spot reactions). E- electrophoretic mobility: 1000 V, 2 hr., 2 N acetic acid. Optical rotation  $\left[ \propto \right]_{D}^{20}$  were measured in DMF. IR spectra (KBr,  $\mathcal{V}_{\text{max}}$  in cm<sup>-1</sup>) were recorded on a Unicam SP 1200 spectrophotometer, UV spectra

(ethanol,  $\lambda_{max}$  nm (log  $\epsilon$ ) on Unicam SP 8000 spectrophotometer. The NMR spectra in DMSO-D<sub>6</sub> were run in Varian T-60 A spectrophotometer using TMS as the internal standard (chemical shift in  $\delta$  -ppm).

General procedure for synthesis of 4-(N-Pht- or N-Tos-aminoacyl )- aminopyridines (III-XIV)-

N-Phthaloyl- or N-tosylamino acid chloride (0.005 mole) was dissolved in dioxane (20 ml) and added dropwise during 30 min. to a cooled solution ( $-5^{\circ}$ ) of 4- aminopyridine (1, 0.47 g, 0.005 mole) in dioxane (25 ml) containing tri-ethylamine (4 ml). The reaction mixture was stirred for 2 hr at 0° and 3-4 hr at room temperature. At the end of the reaction, solid obtained was filtered, washed with water and recrystallized from methanol, ethanol, water or their mixture. The products (III-XIV) were chromatographically homogeneous (ninhydrin negative spot).

General pocedure for synthesis of 2-(N-Pht- or N-Tos-aminoacyl)- aminopyrimidines (XV-XXIV)-

N-Phthaloyl- or N-Tosylamino acid (0.01 mole) and 2- aminopyrimidine (II, 0.01 mole) were dissolved in dioxane (50 ml). The mixture was cooled to 0-5°, dicyclohexylcarbodiimide (2.4 g) added and the mixture stirred for 1-2 hr at 0° and left 24 hr at room temperature. The dicyclohexylurea was filtered off and 4 drops of gl. acetic acid added and the solution refiltered and the filtrate evaporated in vacuo. The residual solid was recrystallized from methanol, ethanol, water or their mixtures. The materials were chromatographically homogeneous when developed with iodine solution or benzidine (ninhydrin and hydroxamate negative spots).

4-(Gly- or  $\beta$  -Ala) aminopyridines (XXV - XXVI) and 2-(L-Phe- or B-Ala) aminopyrimidines (XXVII - XXVIII)-

Each of 4-(N-Pht-Gly- or β-Ala) aminopyridine (III-IV) or 2-(N-Pht-β-Ala or -L-Phe) aminopyrimidine (XVII or XIX) (0.003 mole) was dissolved in dioxane (25 ml) then treated with 0.5 M hydrazine hydrate in ethanol (13 ml.). The reaction mixture was refluxed for 6 hr. The residue obtained after evaporation of the solvent was treated with 2N HCl (50 ml) for 10 min at 50°C. The reaction mixture was cooled and the insoluble phthalyl-hydrazide filtered off. The filtered was treated with Et<sub>3</sub>N (5 ml) for 30 min at 20°C, then Et<sub>3</sub>N.HCl filtered off and the solvent evaporated in vacuum and the residual material was recrystallized from ethanol. The products (XXV-XXVIII) were chromatographically homogeneous when developed with iodine solution, benzidine and gave a positive ninhydrin reaction.

General procedure for synthesis of 4-(N-Tos- or N-Pht- dipeptidyl)- aminopyridines (XXIX - XXXVI)-

4-(Gly- or  $\dot{\beta}$ -Ala) aminopyridine (XXV - XXVI) (0.003 mole) was dissolved in tetrahydrofuran (50 ml) containing triethylamine (2.5 ml) and the mixture stirred for 30 min., and N-tosyl- or N-phthaloylamino acid (0.003 mole) added. The reaction mixture was cooled to 0-5°, dicyclohexylcarbodiimide (0.9 g) added and the reaction mixture worked up as described for (XV - XXIV). The dipeptides (XXIX-XXXVI) were recrystallized from ethanol - water to be

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homogeneous (PC detection with benzidine) and showed negative ninhydrin and silver nitrate reactions.

### 2 - (N-Tos-L-Val-L-Leu) aminopyrimidine (XXXVIII)-

N-Tos-L-Val-L-Leu- $N_2H_3$  (El-Naggar and Latif 1981) (0.0024 mole) was dissolved in a mixture of acetic acid (4 ml) conc. HCl. (2 ml) and water (25 ml). The mixture was cooled to  $-5^\circ$  and sodium nitrite (0.54 g) in water (6 ml) added to it. The dipeptide azide was extracted with ethyl acetate (85 ml), washed successively with HCl (0.5 N),  $H_2O$ , sodium bicarbonate (3 %), water and dried ( $Na_2SO_4$ ). Compound (XXXVII) was prepared by the addition of the dipeptide azide to a cooled ( $-5^\circ$ ) solution of 2-aminopyrimidine (0.0024 mole) in ethyl acetate (40 ml) and keeping the reaction mixture for 24 hr at 0° and for another 24 hr at room temperature. It was washed successfully with HCl (0.2 N), water, sodium bicarbonate (3 %), water and dried ( $Na_2SO_4$ ). The solvent wa removed and the residual materials recrystallized from methanol. The dipeptide (XXXVII) was found to be homogenous (PC single spot with benzidine) and gave negative ninhydrin test.

### RESULT AND DISCUSSION

For the preparation of 4-(N-Pht- or N-Tos-aminoacyl)- aminopyridine derivatives (III-XIV), N-phthaloyl- or N-tosylamino acid chloride was reacted with 4- aminopyridine (I) in dioxanetriethylamine medium using acid chloride procedure. All the products (III-XIV) were obtained in crystalline form in 40 - 75% yield and all gave chromatographically homogeneous spots. Structures of the synthesized pyridine derivatives (III-XIV) are supported by their IR, UV and NMR spectral data. Their IR spectra generally showed a characteristic bands at: 3340, 3140 (NH, N, CONH), 1660, 1560, 1360 (amide I, II and III), 1690 (>C = 0), 3070, 2960, 2780, 1780 and 1440 (pyridine nucleus), thereby confirming their structures. Their UV spectra showed  $^{3}_{max}$  (log  $\epsilon$ ): 262 (2.55), 256 (2.70) characteristic of the pyridyl chromophore. NMR spectra of compounds (III-XIV) exhibit four pyridyl protones in the range  $\delta$  7.00 to 7.60 and other protons assignable to aromatic and amino acid residues.

Coupling of N-phthaloyl- or N-tosylamino acids with 2- aminopyrimidine (II) in dioxane or THF - Et<sub>3</sub>N medium using DCC procedure gave the desired 2-(N-Pht- or N-Tos-aminoacyl)-aminopyrimidines (XV-XXIV). Alternatively coupling of N-phthaloyl- or N-tosylamino acid clorides with 2- aminopyrimidines in benzene-Et<sub>3</sub>N medium gave the products (XV-XXIV) with the same melting points,  $R_f$  and  $[\alpha]^{20}$  and as those obtained by the DCC procedure. The compounds obtained by the acid chloride method needed several recrystallizations (yield 15-35 %). In general, the DCC method gave pure products with higher yield and hence was preferred. Each of the aminoacyl- aminopyrimidine derivatives (XV-XXIV) has the charateristic absorption of the IR spectrum at: 3340, 3140, 3040 (N, NH, CONH), 2940, 2860, 1570, 1390 (pyrimidine nucleus), 1650, 1560, 1260 (amide I, II and III), 1730 (C = 0) The UV absorption showed maxima at 292 (2.90), 242 (3.85) and 252 (3.90) charateristic of the pyrimidine chromophere.

The NMR spectra of compounds (XV-XXIV) exhibit three pyrimidyl protons in the range of 6 7.15 to 7.85 and other protons assignable to aromatic and amino acid residues.

Treatment of 4-(N-Pht-aminoacyl) aminopyridines or 2-(N-Pht-aminoacyl) aminopyrimidines with 1 molar solution of hydrazine in ethanol under mild reflux afforded compounds (XXV-XXVIII). The time required for completion of the reaction was monitored by TLC. Chromatographic and electrophoretic studies on compounds (XXV-XXVIII) revealed their homogeneity (positive ninhydrin reaction,  $E_{XXV} = 15$  cm,  $E_{XXVII} = 19$  cm,  $E_{XXVII} = 16$  cm,  $E_{XXVII} = 12$  cm,  $E_{XXVII} = 16$  cm,  $E_{XXVIII} = 12$  cm,  $E_{XXVII} = 16$  cm,  $E_{XXVII} = 12$  cm,  $E_{$ 

4-(N-Tos- or N-Pht-dipeptidyl) aminopyridines (XXIX-XXXVI) were successively prepared by coupling of N-Tos- or N-Pht-amino acid with 4-(Gly- or -Ala) aminopyridine (XXV-XXVI) in THF containing Et<sub>3</sub>N and using the DCC method. Most of the dipeptides were easily isolated, purified and recrystallized from the proper solvent. The IR spectra of compounds (XXIX-XXXVI) showed characteristic bands: 3370, 3330, 3040 (NH, N, CONH), 1730 ( )C = 0), 1660, 1580, 1360 (amide I, II and III), 3060, 2960, 2880, 1460 (pyridine moiety) and other bands due to dipeptide and pyridine moieties, thereby supporting their structures. Elemental analysis of (XXIX - XXXVI), UV and PMR spectra were consistent with their structures (Table 1).

Synthesis of 2-(N-Tos-L-Val-L-Leu) aminopyrimidine (XXXVII) was achieved starting from the hydrazide Tos-L-Val-L-Leu-N<sub>2</sub>H<sub>3</sub>, which was converted into the corresponding azide. The azide on coupling with 2-aminopyrimidine (II) furnished the dipeptide (XXXVII), which was isolated and purified in the usual manner (El-Naggar *et al* 1977, 1981). The structure of (XXXVII) was confirmed on the basis of its elemental analysis, chromatographic studies, IR, UV and NMR spectral data.

### Biological screeing results

The biological activities of the synthesized compounds were tested using the hole plate method and filter paper disc method, (Carlson-1948; Eastern, 1944; Irving, 1946 and Vincent et al 1944), ad the results compared with the activity of the starting amino compounds (I and II).

2-(N-Pht-Gly) aminopyrimidine (XV) and the corresponding derivatives of N-Pht-L-Ala (XVI and N-Tos-L-Ala (XXI) were found to be active against Bacillus subtilis and inactive against Bacillus mycoids, Bacillus cereus, Esch. coli, Salmonella typhosa and Penicillum chrysogenum. 2-(N-Tos-L-Val-L-Leu) aminopyrimidine (XXXVII) was found to be active against Bacillus subtilis and Bacillus mycoids only. 2-(N-Tos-β-Ala) aminopyrimidine (XXII) inhibited the growth of Bacillus subtilis and Bacillus cereus, while did not inhibit the growth of Esch. coli, Bacillus mycoids, Salmonella typhosa and Penicillum chrysogenum. 2.-(L-Phe) aminopyrimidine (XXVIII) possesses high antimicrobial activity against Bacillus subtilis and Esch. coli. All the protected 4-(N-Pht- or N-Tos-aminoacyl or -dipeptidyl) aminopyridine (III-XIV and XXIX-XXXVI) were found to be biologically inactive towards all the tested microorganisms. On the other hand, 4-(Gly-or β-Ala) aminopyridines (XXV - XXVI) were found to possess high biological activities against Bacillus subtilis, Bacillus mycoids, Bacillus cereus, Salmonella typhosa, Esch. coli. and Penicillum chrysogenum.

Table 1

Physical data of various 4-(N-Tos or N-Pht-aminoacyl, aminoacyl, N-Tos- or N-Pht-dipeptidyl) aminopyridine and 2-(N-Tos- or N-Pht-aminoacyl, aminoacyl, N-Tos-dipeptidyl) aminopyrimidine Derivatives (III-XXXVII)

Compd (Type)	R	Yield (%)	m.p. (°C)	R <sub>f</sub> (TLC)	[×] <sup>20</sup> 1) (deg.)	Mol. Formula	Elemental analysis, %					
							Calc.			Found		
							С	Н	N	С	Н	N
III-(A)	Pht-Gly-	49	236-238	0.76		C15H11N3O3	64.05	3.90	14.90	64.15	3.93	14.89
IV-(A)	Pht-13- Ala-	58	191-193	0.88	[ <del></del>	$C_{16}H_{13}N_3O_3$	65.08	4.42	14.23	64.99	4.70	14.30
V-(A)	Pht-L-Ala-	52	186-188	0.79	-16.8 (c 3.5)	C <sub>16</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub>	65.08	4.42	14.23	65.09	4.45	14.24
VI-(A)	Pht-DL-Ala	45	166-168	0.74		$C_{16}H_{13}N_3O_3$	65.08	4.42	14.23	65.18	4.80	14.26
VII-(A)	Pht-L-Val	63	219-221	0.90	-22 (c 4.5)	$C_{18}H_{17}N_3O_3$	66.87	5.26	13.00	66.95	5.37	13.10
VIII-(A)	Pht-DL-Phe	65	223-225	0.65	l	C22H12N3O3	71.10	4.58	11.32	71.20	4.58	11.35
IX-(A)	Pht-o-Aba-*	58	250-252	0.70		C20H13N3O3	69.97	3.97	12.24	70.07	3.90	12.50
X-(A)	Tos-Gly-	72	256-258	0.55		C14H15N3O3S	55.08	4.91	13.77	54.99	4.95	13.90
XI-(A)	Tos- B -Ala-	63	88-90	0.62	l ——	C15H17N3O3S	56.42	5.32	13.16	56.46	5.60	13.20
XII-(A)	Tos-L-Ala	64	178-180	0.65	-10 (c 0.6)	C <sub>15</sub> H <sub>17</sub> N <sub>3</sub> O <sub>3</sub> S	56.42	5.32	13.16	56.70	5.39	13.18
XIII-(A)	Tos-L-Val	75	142-144	0.60	-14.5 (c 0.6)	C <sub>17</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> S	58.78	6.05	12.10	58.80	6.23	12.17
XIV-(A)	Tos-o-Aba	51	79-81	0.90	l —	C19H17N3O3S	62.12	4.63	11.44	62.21	4.69	11.48
XV-(B)	Pht-Gly-	45	173-175	0.82	l —	C14H10N4O3	59.57	3.54	19.85	59.60	3.94	19.96
XVI-(B)	Pht-L-Ala-	60	205-207	0.66	+20 (c 5.4)	C <sub>15</sub> H <sub>12</sub> N <sub>4</sub> O <sub>3</sub>	60.81	4.05	18.91	60.86	4.11	18.90
XVII-(B)	Pht-B-Ala	41	180-182	0.70	l —	C <sub>15</sub> H <sub>12</sub> N <sub>4</sub> O <sub>3</sub>	60.81	4.05	18.91	60.85	4.21	18.96
XVIII-(B)	Pht-L-Leu	53	210-212	0.79	- 15.5 (c 5)	C <sub>18</sub> H <sub>19</sub> N <sub>4</sub> O <sub>3</sub>	63.71	5.60	16.51	63.82	5.63	16.54
XIX-(B)	Pht-L-Phe	43	150-152	0.55	+ 11 (c 4.7)	C21H16N4O3	67.74	4.30	15.00	67.80	4.38	15.20
XX-(B)	Tos-Glv-	58	160-162	0.93	l ——	C <sub>13</sub> H <sub>14</sub> N <sub>4</sub> O <sub>3</sub> S	50.98	4.57	18.30	50.97	4.58	18.29
XXI-(B)	Tos-L-Ala	48	170-172	0.65	+ 18.9 (c 5)	C14H16N4O3S	52.50	5:00	17.50	52.54	4.98	17.50
XXII-(B)	Tos-B-Ala-	50	198-200	0.85	l —	C14H16N4O3S	52.50	5.00	17.50	52.33	5.11	17.45
XXIII-(B)	Tos-DL-Val-	58	175-177	0.54	l —	C <sub>16</sub> H <sub>20</sub> N <sub>4</sub> O <sub>3</sub> S	55.17	5.74	16.09	55.31	5.72	16.20
XXIV-(B)	Tos-L-Phe-	65	191-193	0.70	+ 15.5 (c 6)	C <sub>20</sub> H <sub>20</sub> N <sub>4</sub> O <sub>3</sub> S	60.60	5.05	14.14	60.58	5.10	14.22
XXV-(A)	Gly-	67	202-204	0.50		C <sub>7</sub> H <sub>10</sub> N <sub>3</sub> OCl	44.80	5.33	22.40	45.01	5.41	22.53
XXVI-(A)	B.Ala-	61	90-92	0.54		C <sub>8</sub> H <sub>12</sub> N <sub>3</sub> OC1	47.60	5.95	20.84	47.69	6.01	20.98
XXVII-(B)	B-Ala-	90	175-177	0.43		$C_7H_{10}N_4O_4$	50.60	6.02	33.73	50.70	6.29	33.80
XXVIII-(B)	L-Phe-	82	168-170	0.74	- 12.5 (c 5.6)	C <sub>13</sub> H <sub>15</sub> N <sub>4</sub> O	64.19	6.17	23.04	64.31	6.23	23.21
XXIX-(A)	Pht-Gly-Gly-	52	161-163	0.75		C <sub>17</sub> H <sub>14</sub> N <sub>4</sub> O <sub>4</sub>	60.36	4.14	16.57	60.34	4.21	16.63
XXX-(A)	Pht-Gly-6-Ala	43	165-167	0.97		$C_{18}H_{16}N_4O_4$	61.36	4.55	15.90	61.40	4.62	16.03
XXXI-(A)	Pht-o-Aba-Gly	49	202-204	0.83		$C_{22}H_{16}N_4O_4$	66.00	4.00	14.00	66.09	4.30	14.12
XXXII-(A)	Pht-o-AbaAla	53	222-224	0.60	· ——	$C_{23}H_{18}N_4O_4$	66.76	4.35	13.53	66.86	4.42	13.61
XXXIII-(A)	Tos-18 - Ala-Gly	73	114-116	0.77		$C_{17}H_{20}N_4O_4S$	54.26	5.32	14.89	54.31	5.34	14.95
XXXIV-(A)	Tos-L-Leu-Gly	61	116-118	0.65	- 20.5 (c 0.6)	$C_{20}H_{20}N_4O_4S$	57.42	6.22	13.59	57.50	6.40	13.60
XXXV-(A)	Tos- B-Ala- B-Ala-	72	120-122	0.80		$C_{18}H_{22}N_4O_4S$	55.38	5.64	14.36	55.42	5.71	14.41
XXXVI-(A)	Tos-DL-Ala-B-Ala-	78	128-130	0.87		C <sub>18</sub> H <sub>22</sub> N <sub>4</sub> O <sub>4</sub> S	55.38	5.64	14.36	55.45	5.72	14.45
XXXVII-(B)	Tos-L-Val-L-Leu-	48	180-182	0.82	- 20.5 (c 5.5)	C22H31N5O4S	57.26	6.27	15.18	57.35	6.47	15.44

<sup>\*)</sup> o - Aba = Ortho-Aminobenzoic acid residue

### 4-(Aminoacyl) aminopyridines and 2-(Aminoacyl) aminopyrimidine Derivatives.

The present investigation reveals that introduction of N-Pht- or N-Tos-aminoacyl, aminoacyl, or N-Tos-dipeptidyl moieties in combination with 2- aminopyrimidine residue induces high and specific biological properties in 2-(N-Pht- or N-Tos-aminoacyl, aminoacyl or N-Tos-dipeptidyl) aminopyrimidines. However, in 4- aminopyridine derivatives blocking of the N-terminal amino group of the aminoacyl moiety with N-phthaloyl- or N-tosyl group results in biologically inactive compounds. In general, the unprotected 2-(aminoacyl) aminopyrimidine and 4-(aminoacyl) aminopyridine derivatives possess the highest antibacterial properties. Other pharmacological studies are in progress.

### CHEMICAL STRUCTURES OF COMPOUNDS (III-XXXVII):

$$\begin{array}{c|c}
 & A \\
\hline
 & N - CH - CO - NH \\
\hline
 & R
\end{array}$$

Compounds III - IX and XXIX - XXXII

III, A = Gly

IV,  $A = \beta$ -Ala

V, A = L-Ala

VI, A = DL-Ala

VII. A = L - Val

VIII, A = DL-Phe

IX, A = O-Aba

XXIX, A = Glv - Glv

XXX,  $A = Glv - \beta - Ala$ 

XXXI, A = O - Aba - Gly

XXXII,  $A = O - Aba - \beta - Ala$ 

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$$CH_3$$
  $\sim$   $SO_2NH - CH - CONH -  $\sim$   $N$$ 

Compounds X - XIV and XXXIII - XXXVI

X, A = Gly

XI,  $A = \beta$  - Ala

XII, A = L - Ala

XIII, A = L - Val

XIV, A = O - Aba

XXXIII,  $A = \beta$  - Ala - Gly

XXXIV, A = L - Leu - Gly

XXXV,  $A = \beta - Ala - \beta - Ala$ 

XXXVI,  $A = DL - Ala - \beta - Ala$ 

(XXV - XXVI)

XXV, A = Gly

XXVI,  $A = \beta$  - Ala

(XV - XIX)

4-(Aminoacyl) aminopyridines and 2-(Aminoacyl) aminopyrimidine Derivatives.

XV, 
$$A = Gly$$
  
XVI,  $A = L - Ala$   
XVII  $A = \beta - Ala$   
XVIII,  $A = L - Leu$   
XIX,  $A = L - Phe$ 

(XX - XXIV)

XX, A = Gly

XXI, A = L - Ala

XXII,  $A = \beta$  - Ala

XXIII, A = DL - Val

XXIV, A = L - Phe

XXXVII, A = L - Val - L - Leu

XXVII,  $A = \beta - Ala$ XXVIII, A = L - Phe

(Compounds Type A)

(Compounds Type B)

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$$HC_3 - O_2NH - CH CONH N$$

(XX - XXIV)

$$\begin{array}{c|c}
A & N \\
\hline
H_2N - CH - CONH \\
R & (XXVII - XXVIII)
\end{array}$$

Compounds III - IX and XXI - XXXII

$$CH_3$$
  $\longrightarrow$   $SO_2$  -  $NH$   $p$   $CH$  -  $CONH$   $\bigcirc$   $N$ 

Compounds X - XIV and XXXIII - XXXVI

4-(Aminoacyl) aminopyridines and 2-(Aminoacyl) aminopyrimidine Derivatives.

$$\begin{array}{c}
A \\
H_2N - CH - CONH - N \\
R
\end{array}$$
(XXV - XXVI)

### REFERENCES

- Abramovitch, R.A. 1974. "The chemistry of heterocyclic compounds", vol. 14, "Pyridine and its derivatives", Interscience, John Wiley, New York, 41-52.
- Brown, D.J. 1970. "The chemistry of heterocyclic compounds," vol. 16, "The Pyrimidines", John Wiley, New York, 230-242.
- Carlson, H.J. 1948. Determination of the antimicrobial activity of different compounds using the hole plate method, J. Bacteriol., 55; 607-612.
- El-Naggar, A.M. and Zaher, M.R. 1977. Synthesis of some peptides containing hippuric acid, urea, thiourea and 2-aminopyridine, Indian J. Chem., 15B; 863-865.
- El-Naggar, A.M., Ismail, I.M. and Gommaa, A.M. 1977. Synthesis of some 2-(N-tosyl or N-phthalylaminoacyl) aminopyridines and some aminothiazole derivatives, Indian. J. Chem., 15B; 850-853.
- El-Naggar, A.M., Ahmed, F.S.M. and Badie, M.F. 1981. Synthesis of some 2-(N-protected of free aminoacyl) aminophenazines and aminophenazone derivatives, J. Heterocyclic Chem., 18; 91-95.
- Epstein, J.A. 1944. Applications and modification of the filter paper disc method, Lab. Clin. Med., 29; 319-325.
- Irving, G.W. 1946. Filamentous mold fungi as test organism, J. Bacteriol., 52; 10-18.
- Vincent, J.C. and Vincent, H.W. 1944. A new method for the determination of antimicrobial properties. The filter paper disc method, Pract. Exptl. Biol., 55; 162-167.

# ٤ - أمينو اسيل أمينو بيريدين و٢ - أمينو اسيل امينو بيريدين التحضير والنشاط البيولوجي لبعض المشتقات الجديدة لمكبات

## فايق سعيد محمد أحمد

أحمد محمد النجار

حسن عبد البارى

محسن سعيد عبد المنعم عبد السلام

طريقة الكلوريد الحامضي والكاربودايميد . ومعالجة مركبات ٤ - ( ن - فثاليل جلاسيل أو الانيل ) امينو بيريدين ومشتقات ٢ - ( ن - فثاليل فينيل الانيل او بيتا الانيل ) امينو بيريدين في الايثانول امينو بيريدين و٢ - ( ن - توزيل أو ن - فثاليل - امينو اسيل ) امينو بيريدين وذلك باستخدام تضمن البحث تخليق مجموعة جديدة من مركبات ٤ - (ن - توزيل أون - فثاليل - امينو اسيل) نتجت المشتقات الطليقة المحتوية على مجموعات الامين الغير محمية .

وشمل البحث على تخليق مجموعة من مركبات ٤ - ( ن - فثاليل او ن - توزيل - بيتيد ثنائي ) امينو بيريدين وذلك باستخدام طريقة الكاريودايميد ومركب ٢ - ( ن - توزيل فثاليل ـ ليوسيل ) امينو بيريدين باستخدام طريقة الازيد

وبدراسة النشاط البيولوجي للمركبات التي تم تخليقها اتضح ان عدد ثمانية مركبات ذات نشاط بيولوجي عال تجاه مختلف الكائنات الدقيقة .