

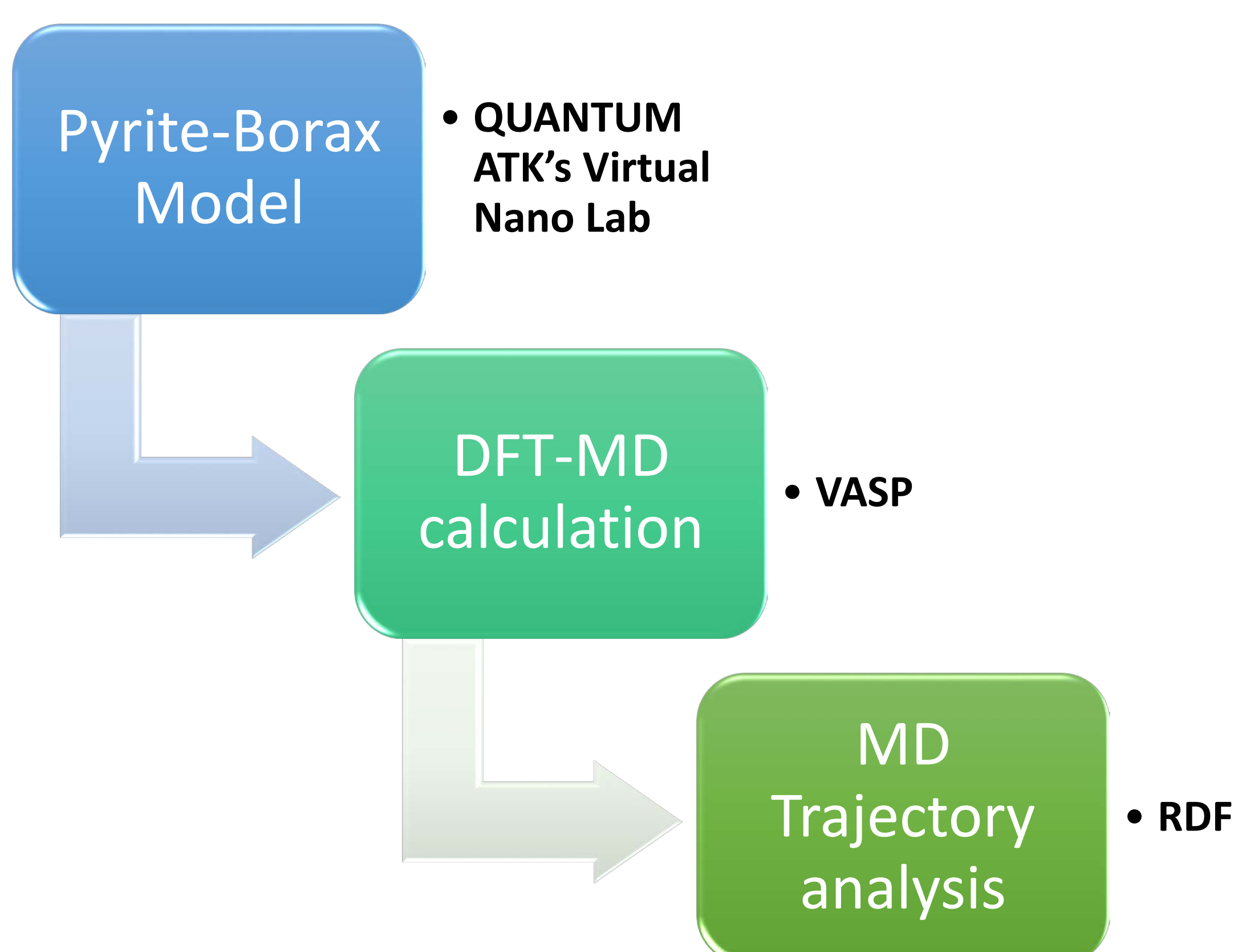
Borax aids pyrite scale dissolution leading to scale removal in oil and gas pipelines.

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BACKGROUND

Iron sulfide form scales such as pyrite in pipes which lead to loss of production in the oil and gas industry. Computational tools such as DFT-MD (Density Functional Theory-Molecular Dynamics) may be used to understand and model chemical compounds that would be effective in removing these scales.

METHODS



RESULTS

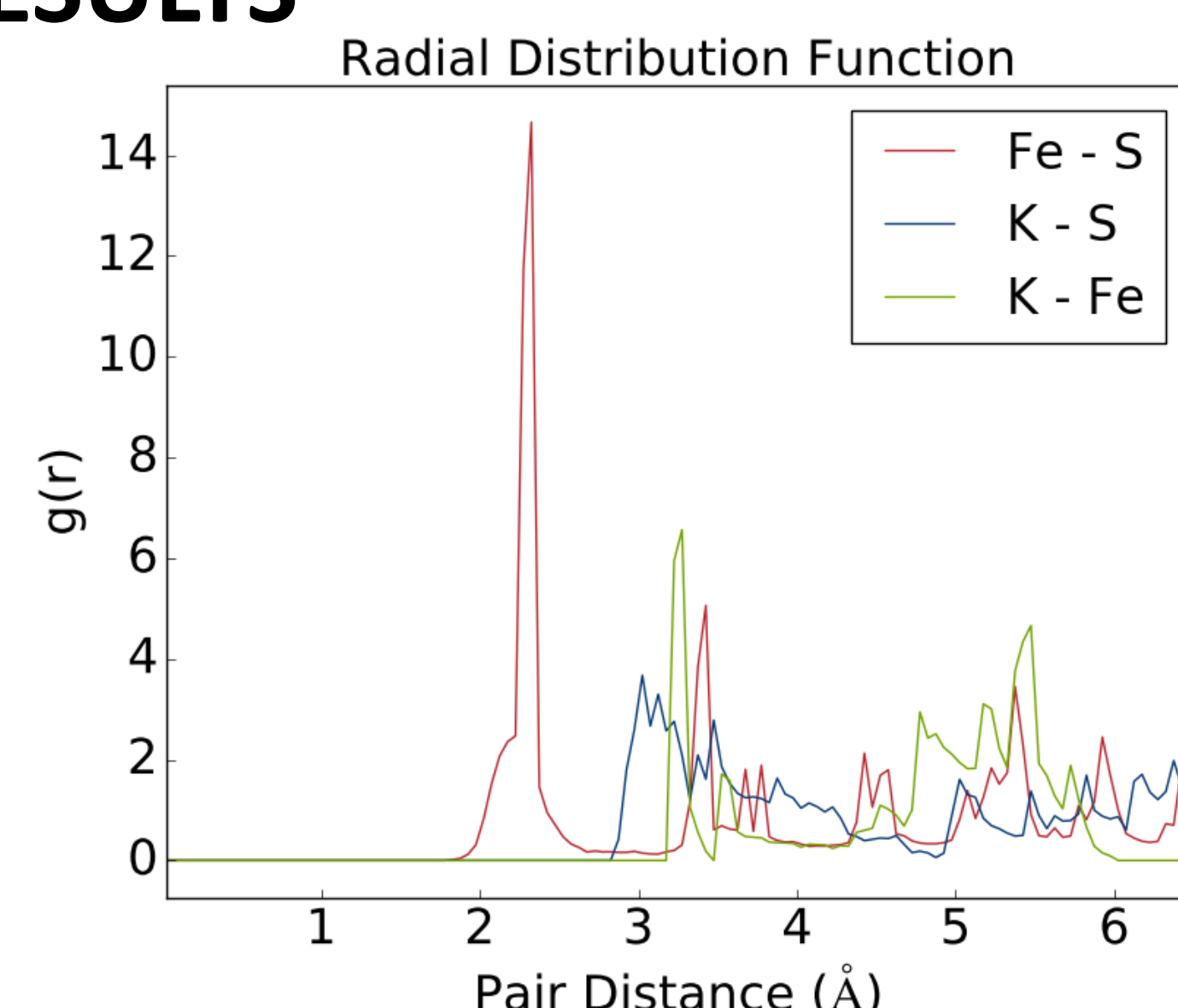


Fig 1. The RDF of the pyrite-borax system for Fe-S, K-S, and K-Fe bonds

- The potassium ion (K^+) form bonds with the sulfur (S) atoms in the top layer of the pyrite rather than iron (Fe) as shown in the radial distribution function (RDF), and geometrical structures.
- K^+ present in borax is the predominant factor that accounts for pyrite dissolution.
- The absence of chelating agent may lead to the formation of Fe-S bonds therefore, restoring the system to its initial state.

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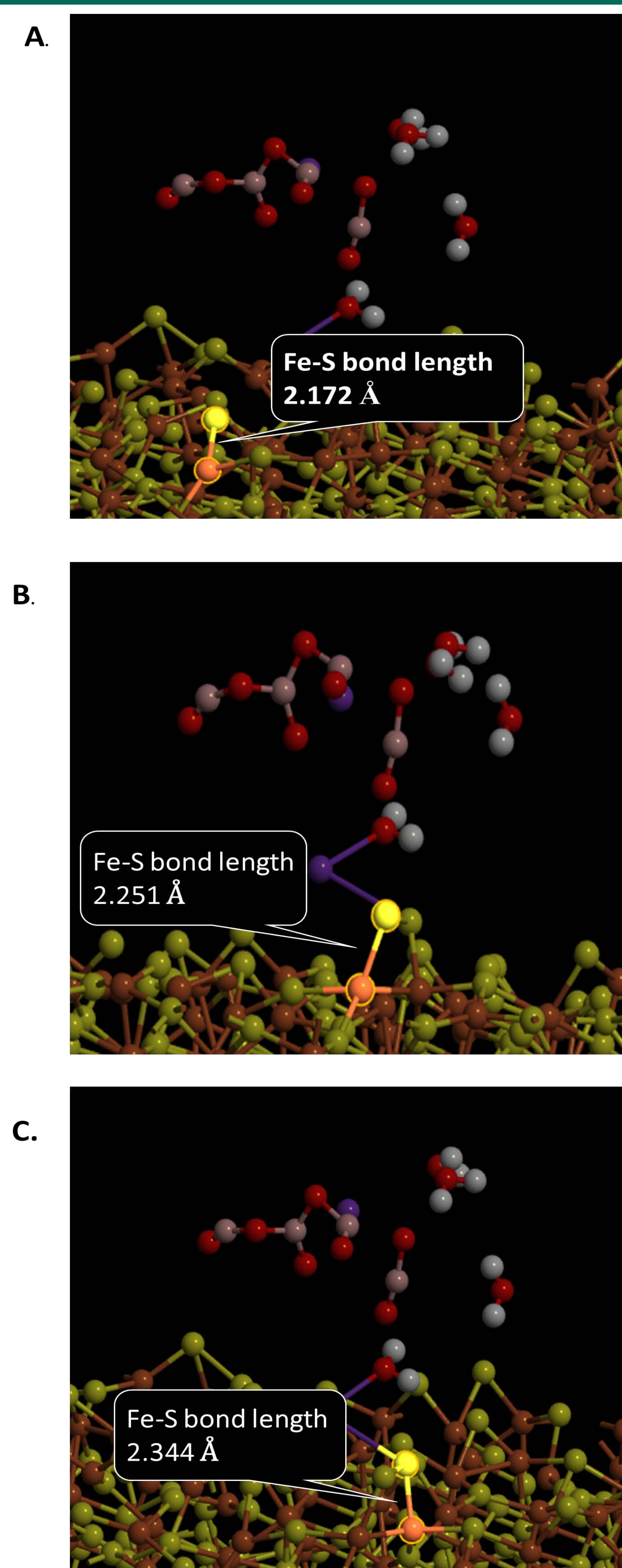
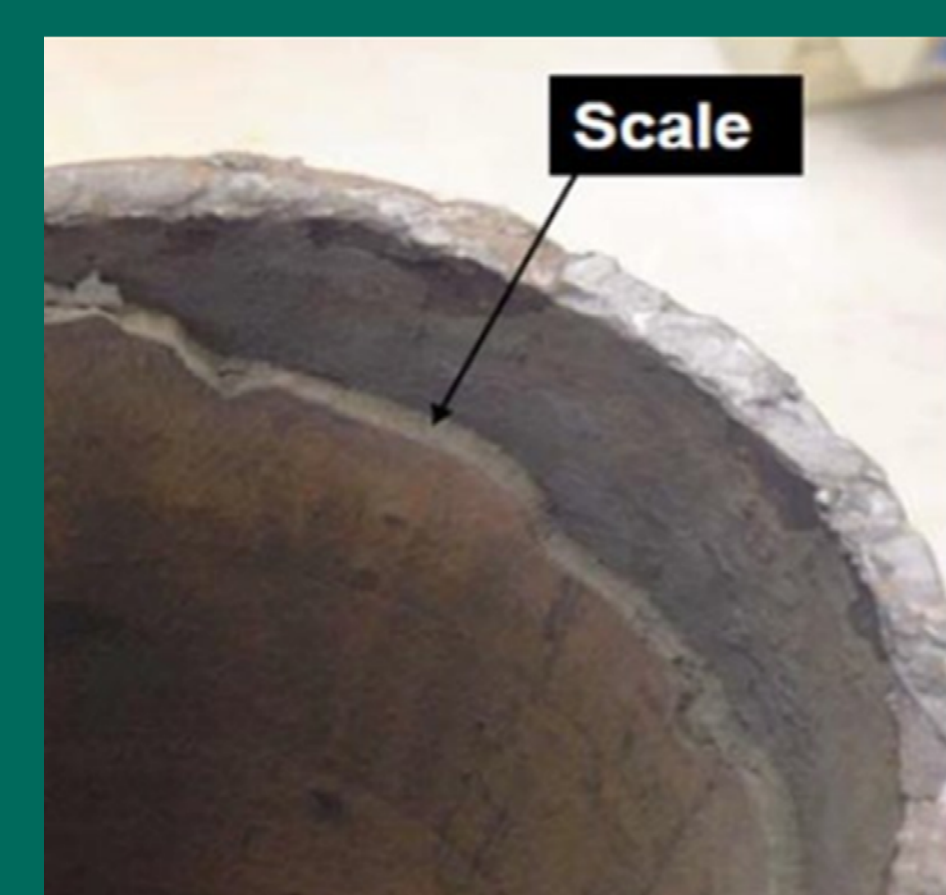
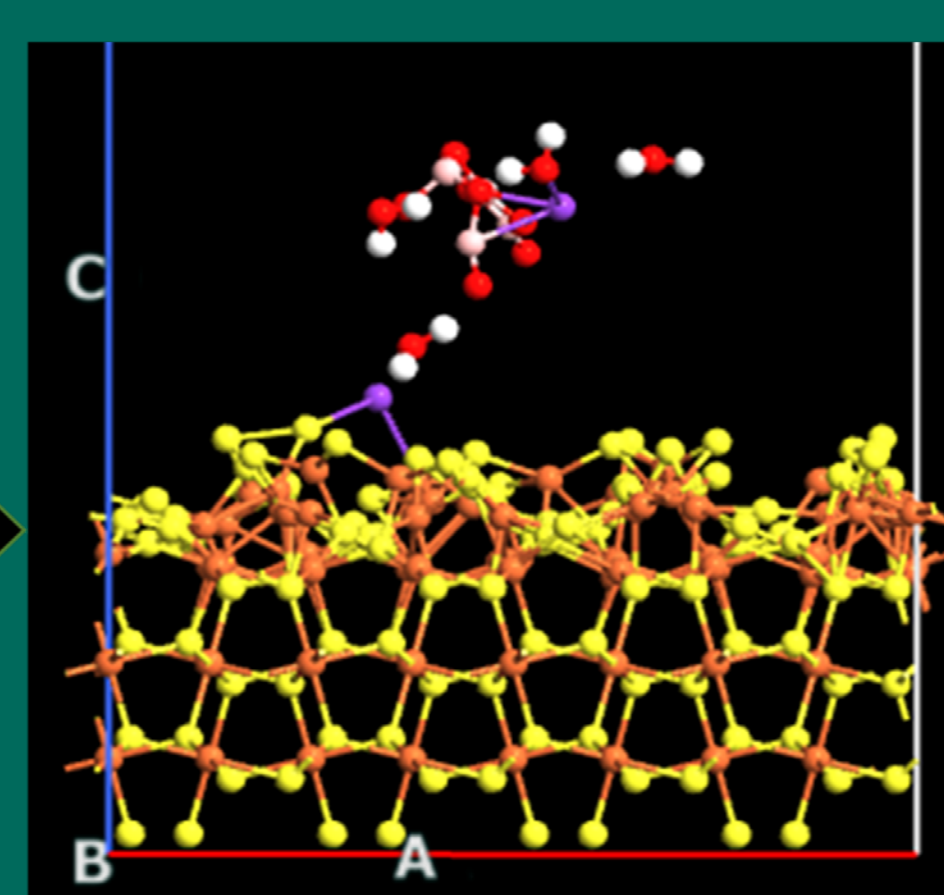


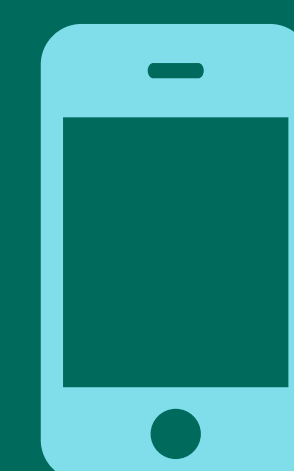
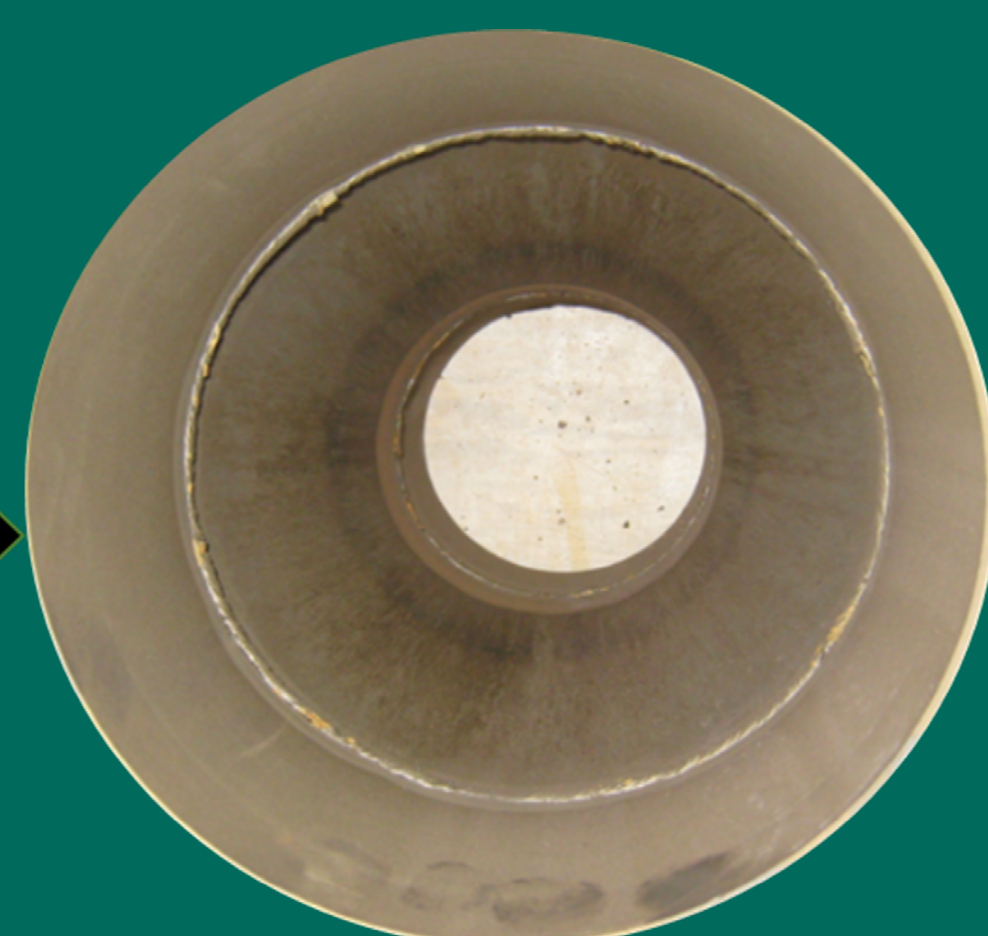
Fig 2. Difference in bond lengths of Fe-S bonds depicting that Sulfur atoms are leaving the surface



DFT-MD



Scale Removed



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