

Curcumin gum Arabic nanoparticles demonstrate potent antioxidant and cytotoxic properties in human cancer cells

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Abstract: The main purpose of the study was to enhance the stability and therapeutic effects of Curcumin (Cur) through nanoformulation with gum Arabic (GA) as a coating agent through an efficient synthetic approach. The antioxidant properties of the developed nanoparticles (Cur/GANPs) were assessed through several *in vitro* assays, such as β -carotene bleaching activity, DPPH, and nitric oxide scavenging activities in addition to evaluating its inhibitory activity on angiotensin-converting enzyme (ACE). The cytotoxicity of Cur/GANPs was evaluated *in vitro* using different types of human cancer cells including breast cancer (MCF7, MDA-MB231), liver cancer (HepG2), and colon cancer (HT29) cells. The prepared particles displayed an elliptical shape with a size ranging between 20–260 nm and a potential difference of –15 mV. The Cur/GANPs exhibited significant antioxidant activity compared to free curcumin when using concentrations between 31.5 and 500 μ g/mL. The Cur/GANPs also had inhibited the growth of all cancer cell lines in a proportional trend with concentrations used. Hence, the encapsulation with gum Arabic has augmented the antioxidant and anti-neoplastic effects of Curcumin. Therefore, Cur/GANPs may have effective therapeutic properties in diseases attributed to oxidative stress like cancer and hypertension.

Abbreviations

GA:	Gum Arabic
Cur/GANPs:	Curcumin Gum Arabic Nanoparticles
Cur:	Curcumin
ACE:	Angiotensin-converting enzyme
NF-KB:	Nuclear Factor kappa-light-chain-enhancer of activated B cells

Introduction

Oxidative stress is a condition that contributes to various deleterious changes in the human body as a part of the aging process or in pathological conditions like cancer and

cardiovascular diseases (Reckziegel *et al.*, 2016). Such condition results from the accumulation of oxidative radicals to critically high levels exceeding the antioxidant capacity of the body leading to adverse consequences. The oxidative radicals of oxygen include hydrogen peroxide and superoxide radicals which instigate pathological changes in the form of DNA and protein oxidation and lipid peroxidation (Hanahan and Weinberg, 2011). The impact of reactive oxygen species on the cellular DNA contributes significantly to the neoplastic transformation of cells that is characterized by the uncontrolled proliferation with the other properties of cancer cells (Prasad and Tyagi, 2017). Therefore, the use of therapeutic antioxidants provides a promising tool that may deter adverse outcomes such as cancer or cardiovascular diseases.

Curcumin, the yellow polyphenol extract of turmeric, with a wide range of therapeutic uses due to its antioxidant, antineoplastic, and anti-inflammatory properties (Gangwar *et al.*, 2013; Chidambaram and Krishnasamy, 2014; Gupta *et al.*, 2015;

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Amalraj *et al.*, 2017; Fadus *et al.*, 2017). The anticancer activities of curcumin are mostly attributed to its efficiency in reverting protein oxidation including protein kinase, tyrosine kinase, and NF-kb (Notarbartolo *et al.*, 2005; Mondal *et al.*, 2013; Majumder *et al.*, 2014; Meiyanto *et al.*, 2014). However, these therapeutic attributes are challenged by poor solubility, rapid elimination, and low bioavailability (Yin *et al.*, 2013; Meiyanto *et al.*, 2014).

The nanomedicine field had contributed significantly to improving the potency, lowering the toxicity margin of treatments in diseases like cancer (Hussein *et al.*, 2012; Ibrahim *et al.*, 2020). The applications may include increasing the precision of treatments such as radiotherapy and phototherapy (Ardakani *et al.*, 2020; Meidanchi, 2020a; Meidanchi and Ansari, 2020; Meidanchi and Motamed, 2020). The use of nanoparticles with their unique properties may also help in overcoming the low solubility of hydrophobic agents like curcumin (Wang *et al.*, 2013; Jafarzadeha *et al.*, 2019).

The use of amphiphilic polymer coating may enhance the stability and may impose special surface characteristics on the active agents in the nanoformulation. Hydrophilic natural polymers have been widely used for encapsulation of hydrophobic agents like curcumin to achieve the sought properties. In this context, the use of gum Arabic as a coating material may improve the antioxidant activity of curcumin due to its additive antioxidant effect, biodegradability, biocompatibility, and its chemical properties that may improve the precision of the treatment (Sarika *et al.*, 2014).

Gum Arabic (GA), is a polysaccharide polymer that is naturally produced from the branches and stems of Leguminosae Acacia Senegal and is commonly used in many pharmaceutical and food applications (Butstraen and Salaün, 2014; Bae *et al.*, 2019; Pieczykolan and Kurek, 2019). The highly branched molecular structure of GA induces steric repulsion properties and enhanced the *in vivo* stability of nanoparticles, whereas the carboxyl groups can easily react with bioactive agents (Wu and Chen, 2010). Among its many properties, its strong antioxidant property is the most documented characteristic. As a polymeric nanoparticle with special physical and chemical properties, GA is a promising candidate nanoparticle for various biomedical preparations (Al-Mosawi, 2002; Meidanchi, 2020b; Raikos *et al.*, 2016).

In this study, a novel nanoformulation was improved based on gum arabic to enhance the antineoplastic effects of curcumin (Cur/GANPs). Hence, the Cur/GANPs were formulated using the freeze-drying method. Multiple assays were used to evaluate and validate the antioxidant potential of Cur/GANPs and Cur. The cytotoxicity of both free curcumin and Cur/GANPs was individually determined in different cancer cell lines at various time and dose intervals based on the MTT assay.

Materials and Methods

Materials

Gum Arabic (ENNASR Co., Sudan), Curcumin (Biolutions Resources, China), RAW 264.7, MCF7, HepG2, HT29, and MDA-MB231 (American Type Culture Collection (ATCC), USA). Dexamethasone, 1,1-diphenyl-2-picrylhydrazyl (DPPH), TROLOX, Linoleic acid, Hippuryl-histidyl-leucine sulfonamide, sodium nitrite, Angiotensin-converting enzyme, N-(1-naphthyl) ethylenediamine dihydrochloride and high glucose Dulbecco's

Modified Eagle's (DMEM)(Sigma-Aldrich, Malaysia). β -carotene and Tween 80 (R&M Co., China).

Formulation of Cur/GANPs

Cur/GANPs were prepared based on the freeze-drying method (Abdelwahed *et al.*, 2006; Hassani *et al.*, 2020). Briefly, the gum Arabic solution was produced by adding 0.70 g of GA and dissolving it in 50 mL of distilled water. Then, an aqueous solution of curcumin (1 mg/mL) was dissolved in ethanol and mixed with the GA solution in a ratio of 1:4. The mixture obtained was further to a stirring for 72 h then subjected to a high-pressure homogenizer (HPH) (1000 bar, 8 cycles) and frozen at -80°C . The resulting dispersion was then freeze-dried using a freeze dryer for 24 h at -55°C .

Characterization of Cur/GANPs

The structural properties and crystallinity of Cur/GANPs samples were assessed by the X-ray diffraction (XRD) (Shimadzu refractometer, Philips) with CuK α incident beam. The XRD analysis was carried out at $2\theta = 20^{\circ}$ – 60° and a scan speed of 2° per minute.

The electrical potential and the size distribution of Cur/GANPs were determined in this study using the Zeta Sizer (UK) and UTHSCSA Image software. Transmission electron microscopy (TEM) of Cur/GANPs was conducted using Hitachi H-7100 electron microscope (Japan) to observe the sizes and assess the homogeneity of Cur/GANPs.

Size distribution, zeta potential, and stability of Cur/GANPs

The size distribution and zeta potential of Cur/GANPs was investigated by using a zeta sizer (Malvern Nano-ZS90, London, UK). Then, 0.2 mL of samples were diluted in 0.8 mL of deionized water then subjected to ultrasonic vibration to avoid the phenomenon of aggregation of nanoparticles.

The stability study was assessed based on the changes observed in the zeta potential and size distribution of Cur/GANPs. The stability assay was carried out after 10 days of preparation of Cur/GANPs at three temperatures of storage (-20°C , 4°C , and 25°C). The measurement was performed in triplicate.

Controlled release study of curcumin from GANPs into suitable media by UV-Vis spectrometer

The amount of Cur released from Cur/GANPs in the buffer solutions was recorded based on Perkin Elmer UV-Vis spectrophotometer at pH 7.4 and pH 4.8. The kinetic release was investigated by the addition of 8 mg of Cur/GANPs into the buffer solution under shaking at 37°C . The measure of absorbance was carried out at different times of 0.5, 1.0, 2.0, 3.0, 4.0, 5.0, 6.0, 12.0, 24.0, 48.0, and 72.0 h using spectrophotometer (Shimadzu 1800 UV) at $\lambda_{\text{max}} = 430 \text{ nm}$.

The percentage of the release of Cur from Cur/GANPs into the media was calculated by:

$$\% \text{ Release} = \frac{\text{Concentration of drug at time } t}{\text{Concentration corresponding to 100\% release of drug}} \times 100$$

The concentration that corresponded to 100% release of drug was calculated based on the absorbance measured after 7 days of the release of Cur from Cur/GANPs into the Phosphate buffer saline (PBS) at pH 7.4 and pH 4.8. To

compare the percentages of Cur released from GANPs with that from the physical mixture of Cur and GA, the same weight ratios as the GANPs were prepared to determine the release as described above. The release of active compounds was estimated as a percentage accordingly.

The behavior of Cur/GANPs at pH 7.4 and 4.8

Zeta potential and particle were measured to assess the deformation behavior and aggregation of nanoparticles at pH 7.4 and 4.8.

The Scavenging Activity of Cur/GANPs with DPPH

The DPPH scavenging test is based upon the reducing capacity of DPPH in the presence of antioxidant compounds by donating an electron or hydrogen to the free radicals indicated by the change of purple DPPH radicals into yellow stable compound (Sarika *et al.*, 2015). Due to the potent antioxidant capacity of Trolox, it was used as a positive control in this assay. A serial dilution of 100 µg of Trolox in 1 mL of methanol was used in the range of 50–200 µg/mL. Two hundred microliters of DPPH solution were added to each well of the 96-well microplate. After shaking for 1 min and incubating the samples in a dark environment, the spectrophotometric light absorbance was measured at 517 nm. The scavenging activity was calculated according to the following equation:

$$\text{DPPH activity (\%)} = \left[1 - \frac{A_a - A_b}{A_c} \right] \times 100 \quad [1]$$

where A_a, A_b, and A_c are the absorbance of samples, absorbance recorded without DPPH, and control absorbance, respectively.

The Nitric-Oxide (NO) scavenging activity of Cur/GANPs

The RAW 264.7 cells were cultured with a cell density of 1 × 10⁶ cells/100 µL of cell culture media in each well of the 96-well plates for 24 h. Then, free curcumin and Cur/GANPs were added with concentrations ranging between 15.6 and 500 µg/mL. The cells were then challenged with 10 µg/mL of lipopolysaccharide (LPS) for 20 h. The released Nitric oxide (NO) was indirectly quantified by measuring the quantity of nitrite released into the supernatant of macrophage cells. The nitrite quantification was determined by adding 100 µL of Griess reagent to the medium of each well in the 96-well plate. The Griess reagent consists of 1% (w/v) sulfonamide and 0.1% (w/v) N-(1-naphthyl) ethylenediamine dihydrochloride. The microplate reader was used to evaluate the absorbance at 540 nm. A standard curve was plotted based on various concentrations of sodium nitrite to calculate the concentration of nitrite in the medium (Al-Qubaisi *et al.*, 2013; Ranneh *et al.*, 2016).

β-carotene bleaching assay

The assay is based on the oxidation of linoleic acid indicated by the decrease of the yellow color (bleaching) of β-carotene. This assay was implemented with some modification to the method previously described (Prieto *et al.*, 2012). Briefly, the β-carotene solution was prepared after the dissolution of 2 mg of β-carotene in 10 mL of chloroform then evaporated at 40°C under vacuum. After

evaporation, Tween 80 (400 mg) and linoleic acid (40 mg) were mixed then added to 100 mL of distilled water. The mixture was vigorously shaken for 10 min. An emulsion without β-carotene was prepared and used as blank. After adding free curcumin and Cur/GANPs at the specified concentrations (50 µL) with β-carotene emulsion (200 µL) into each well of the 96-well plate, the mixtures were incubated at 50°C in dark environment. At the end, 470 nm light absorbance was measured at 0 min and then at 20 min intervals for 100 min. The antioxidant activity was evaluated as follows:

$$AA_0 = \left[1 - \frac{(A_0 - A_t)}{(A_{0c} - A_{tc})} \right] * 100 \quad [2]$$

where A₀, A_t, A_{0c}, and A_{tc} are the absorbance values of samples at t = 0, t = 100 min, and controls at t = 0 and t = 100 min respectively. Various concentrations of α-tocopherol were used as standards.

Antihypertensive activity

This assay is based on the inhibition of Angiotensin converting enzyme (ACE) activity *in vitro* indicated by colorimetric evaluation of the conversion of hippuryl-histidyl-leucine to hippuric acid in the presence of an inhibitor (Cur/GANPs). In this assay 100 µL of Cur and Cur/GANPs were mixed with ACE (25 µL, pH 8.3) at 37°C. To assess the ACE inhibition activity, 10 µL of hippuryl-histidyl-leucine (3.5 mM) was added to the mixture then incubated for 30, 60, and 90 min. The reaction was stopped by adding 50 µL of HCL (1M) to the solution followed by extraction of hippuric acid by Ethyl acetate (1 mL). After evaporation at 120°C, the residue was re-dissolved in deionized water. Finally, the amount of hippuric acid was assessed by the measurement of absorbance at 228 nm. The emulsion prepared without curcumin or Cur/GANPs was used as a control (Ahmad *et al.*, 2017).

Cell viability assay

Four human cell lines, human colon cancer cells (HT29), human liver cancer (HepG2), and breast cancer (MDA-MB231 and MCF7) cell lines were used in the cytotoxicity assay. Two hundred microliters of a 1 × 10⁴ cells/mL suspension were seeded into each well of the 96-well plate. After incubation for 24 h, the media was aspirated, and the cells were treated with fresh media (200 µL) including free curcumin and Cur/GANPs of various concentrations (15.6 to 100 µg/mL). The plates were then incubated with the treatments for 72 h at 37°C with 5% CO₂. The cell viability was determined based on the MTT assay in which the media was removed, and the wells were washed three times using PBS to ensure that curcumin and Cur/GANPs were removed. Then twenty microliters of MTT solution (5 µg/mL) were added into each well of the 96-well with fresh media then mixed gently and incubated for 4 hours at 37°C with 5% CO₂. Then MTT-including culture medium was replaced by 200 µL/well of DMSO to dissolve the formazan crystals. The absorbance was determined at 570 nm and the amount of drug required to inhibit 50% of cell growth (IC₅₀) was calculated from the dose-response patterns from each cell line and compound (Al-Qubaisi *et al.*, 2013).

Statistical Analysis

The experiment was carried out in triplicates with multiple replicates. The data were used to calculate mean values and standard deviation. The comparison of the mean values was analyzed by the SPSS 16 for windows software. The difference of measurements was evaluated using one-way analysis of variance (ANOVA) and independent *t*-test with $p < 0.05$ considered statistically significant.

Results

In the X-ray diffractogram of curcumin, sharp diffraction peaks were demonstrated at different diffraction angles 25.94°, 27.96°, 34.32°, and 37.82° as observed in section d of Fig. 1. Sections a, b and c demonstrate the diffractogram of

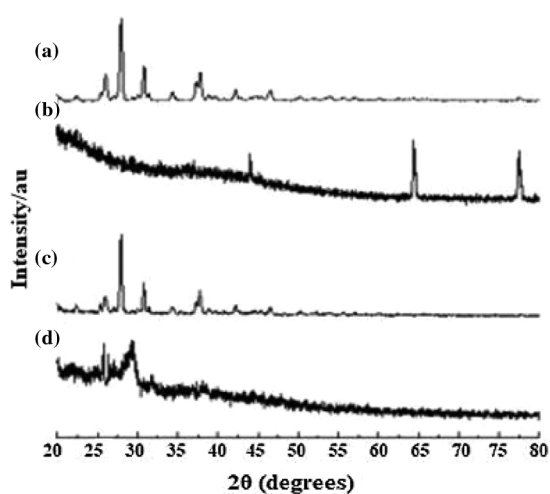


FIGURE 1. X-ray diffractogram: (a) curcumin, (b) GA, (c) physical mixture, (d) Cur/GANPs nanoparticles.

(a) curcumin, (b) GA, (c) physical mixture of Cu and GA, (d) Cur/GANPs nanoparticles.

Cur/GANPs have been formulated and the clear solution of Cur/GANPs displayed a particle size over the range of 20–260 nm (Polydispersity Index PDI = 0.32 ± 0.13), with the potential of -24 mV as displayed in Fig. 2a.

At freezing temperature of -20°C , the Cur/GANPs revealed a tendency to aggregate after melting that was confirmed by the reduction of the negative potential (Fig. 3A) and the increase of particle size of Cur/GANPs (Fig. 3B). The stability of the nanoparticle's formulation was also assessed in acidic and alkaline conditions as demonstrated (Fig. 4). aggregation and deformations were noticed in acidic pH as indicated with increased size and potential.

The *in vitro* drug release in the GANPs was calculated based on the curve of the UV-Vis spectrometer at λ_{max} , 430 nm for curcumin as shown there was a steady release of curcumin from the nanoparticle formulation over 2 days in acidic and alkaline pH (Fig. 5A). while the physical mixture had an immediate release pattern as observed in (Fig. 5B).

The Cur and Cur/GANPs scavenging of free radicals were evaluated using different *in vitro* assays including the DPPH, LPS stimulation, and the β -carotene bleaching assays. Trolox was utilized as a positive control due to its capacity to dissolve in the aqueous system and had shown a strong scavenging activity against DPPH. Upon the evaluation of curcumin and Cur/GANPs, both showed an antioxidant activity at the tested concentrations in the range 50–200 $\mu\text{g}/\text{mL}$ as illustrated in Fig. 6. As demonstrated, the percentage of DPPH scavenging activities of Cur/GANPs was higher than free curcumin in each of the concentrations used ($p < 0.05$) as shown in Fig. 8.

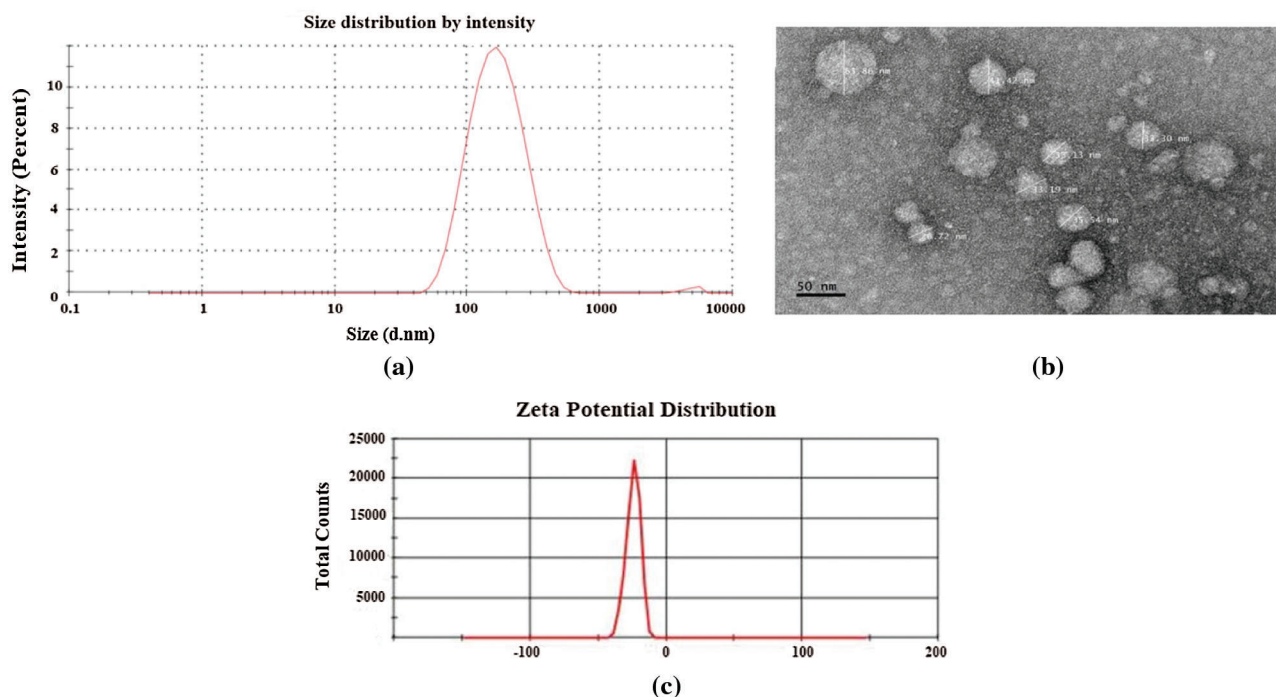


FIGURE 2. Estimation of Cur/GANPs size using Zeta sizer (a), Transmission electron microscope image of the developed NPs (b), and the Zeta potential (c) of Cur/GANPs.

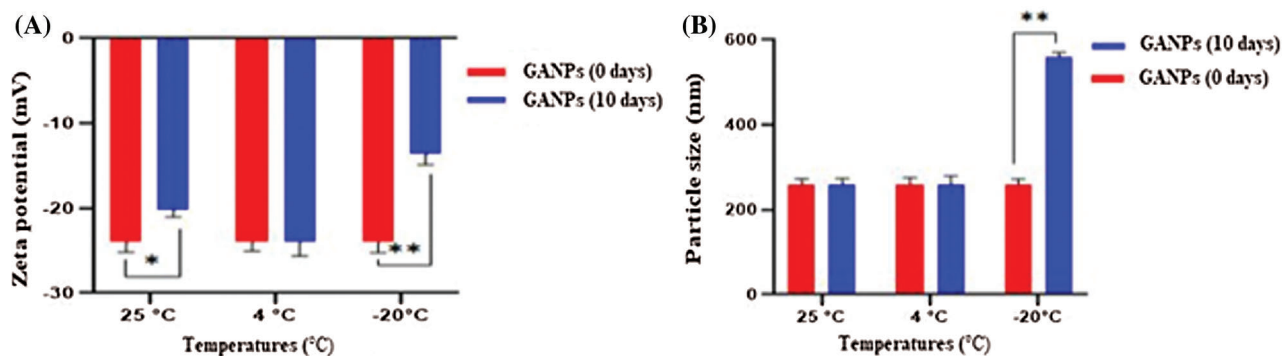


FIGURE 3. The stability of GANPs suspensions at three temperature settings of storage. The indicators of stability included (A) Zeta potential, and (B) Particle size. The data presented as Mean \pm SD of three experiments (n = 3). * $p < 0.05$, ** $p < 0.01$.

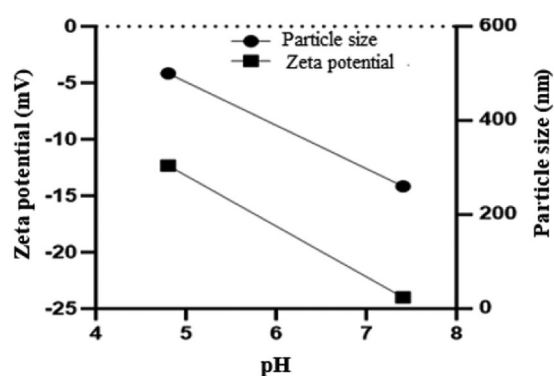


FIGURE 4. Aggregations and deformation of Cur/GANPs at pH 4.8 and 7.4. Mean \pm SD of three experiments (n = 3).

Another assay was used to determine the antioxidant and anti-inflammatory properties of treatments known as the LPS stimulation test which was experimented with in the murine macrophage cell line (RAW 264.7). The murine macrophages treated with curcumin and Cur/GANPs revealed a considerable inhibition of NO production in their media as shown in Fig. 7A. When cells were pre-treated with curcumin or Cur/GANPs, LPS stimulation of nitrite oxide production was inhibited in a concentration-dependent manner. As shown, there was a significant increase in the viability of RAW 264.7 cells Fig. 7B. Cur/GANPs scavenging activity was twice higher than free curcumin at a concentration ranging from 31.5 to 500 $\mu\text{g/mL}$ ($p < 0.05$) as shown in.

The antioxidant activity of β -carotene increased in a dose-dependent manner with Cur/GANPs treatment as shown in Fig. 9. The highest antioxidant activity (70.11%)

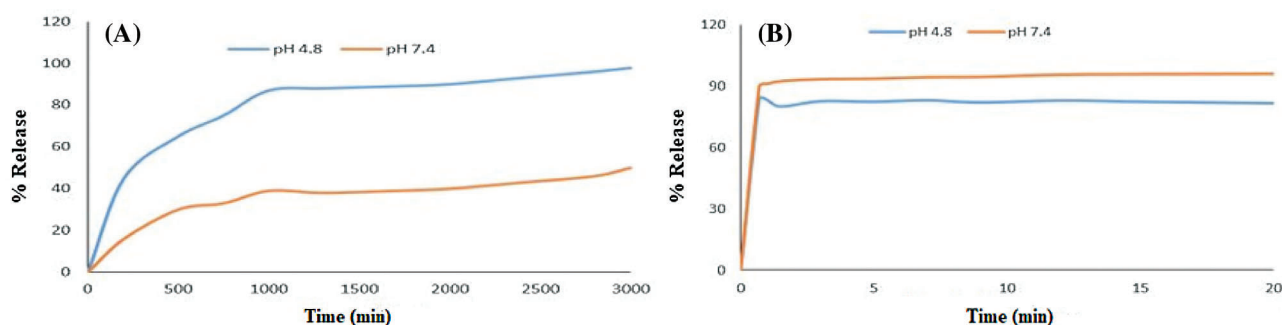


FIGURE 5. Release profiles of Cur from (A) GANPs, (B) physical mixture at pH 4.8 and pH 7.4. Mean \pm SD of three experiments (n = 3).

was reported at 500 $\mu\text{g/mL}$ concentration of Cur/GANPs that was significantly higher than the other concentrations (p -value < 0.05).

ACE inhibitors are medications prescribed to treat high blood pressure. In this study, the antihypertensive activity of Cur and Cur/GANPs was evaluated by converting hippuryl-histidyl-leucine to hippuric acid (Hussein *et al.*, 2012). Cur/GANPs exhibited a more potent inhibitory activity than the free curcumin ($p < 0.05$) suggesting a stronger antihypertensive activity as shown in Fig. 10.

In vitro cytotoxicity tests in cancer cells help as a preliminary screening tool of anticancer properties before more specific *in vitro* and *in vivo* assays are applied. The cytotoxic properties of curcumin in the free form and the nanoparticles formulation (Cur/GANPs) were assessed in different human cancer cell lines based on the reduction of MTT reagent by cancer cells as shown in Fig. 11. As shown,

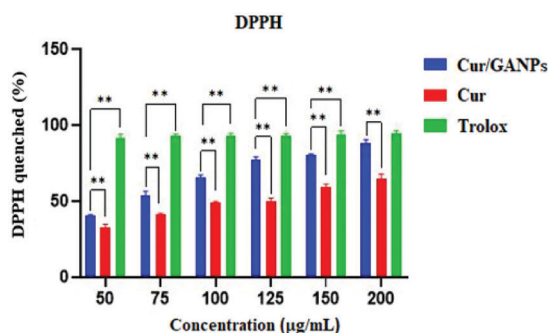


FIGURE 6. DPPH scavenging of free curcumin and Cur/GANPs. Independent t -test, Data shown are mean value \pm SD, independent t -test, * p -value ≤ 0.05 , ** p -value ≤ 0.01 .

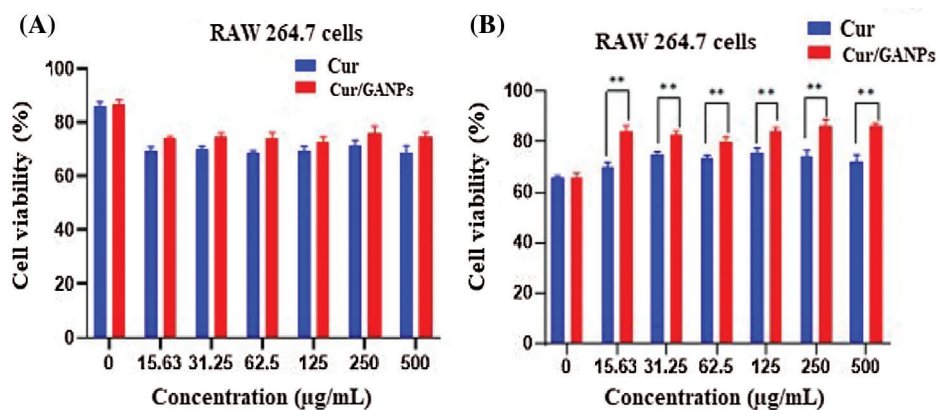


FIGURE 7. Cytotoxicity assay in RAW 264.7 cells treated with free curcumin and Cur/GANPs for 24 h at various concentrations without LPS stimulation (A) and with LPS stimulation (B); Independent *t*-test, Data shown are mean value \pm SD, * *p*-value \leq 0.05, ** *p*-value \leq 0.01.

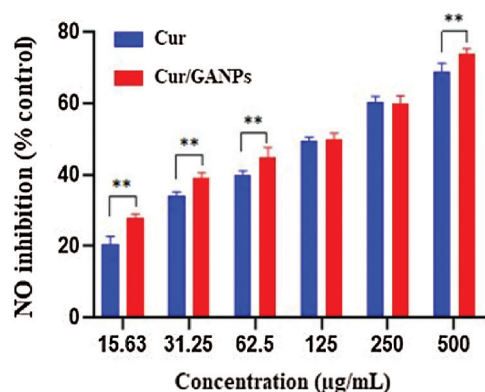


FIGURE 8. Nitric oxide radical scavenging activities of RAW 264.7 cells treated with different concentrations of free curcumin and Cur/GANPs. Independent *t*-test, Data shown are mean value \pm SD, independent *t*-test, * *p*-value \leq 0.05, ** *p*-value \leq 0.01.

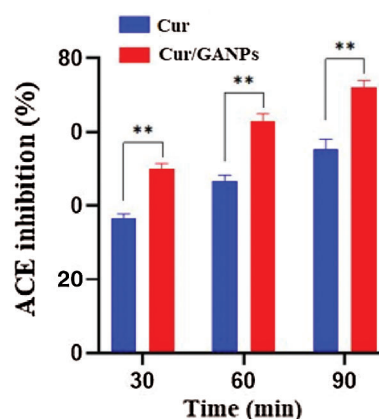


FIGURE 10. ACE inhibition (%) for free curcumin and Cur/GANPs after 30, 60, and 90 minutes. Independent *t*-test, Data shown are mean value \pm SD, independent *t*-test, * *p*-value \leq 0.05, ** *p*-value \leq 0.01.

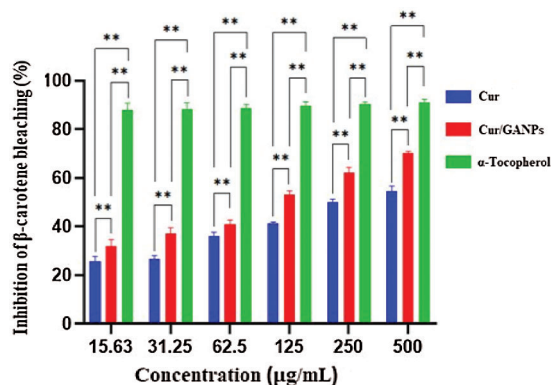


FIGURE 9. Inhibition (%) of β -carotene bleaching with free curcumin and Cur/GANPs. Independent *t*-test, Data shown are mean value \pm SD, independent *t*-test, * *p*-value \leq 0.05, ** *p*-value \leq 0.01.

there was a dose-dependent inhibition of cell viability with Cur/GANPs and the IC_{50} values of Cur/GANPs were significantly lower than free curcumin in all cancer cell types (HepG2, MCF-7, HT29, and MDA-MB231), indicating better therapeutic properties (Table 1).

Discussion

The freeze-drying method was selected in the preparation of the nanoparticles among the other methodologies due to its undemanding and eco-friendly characteristics. The

produced freeze-dried powder was further reduced to ultrafine powder. The cryo-chemical method used in this study has provided several advantages compared with other conventional methods such as the small size of the end product and the greater purity and homogeneity.

The XRD diffractogram illustrated the crystalline nature of curcumin (Fig. 1). The same diffraction peaks appeared in the physical mixture pattern with a reduction of intensity. The GA polymer didn't show obvious peaks at 64.15° and 77.35° indicating the amorphous state of GA. Due to the interaction between curcumin and GA, Cur/GANPs did not exhibit the same diffraction peaks of free curcumin where the major peaks totally vanished. This suggests the conversion of curcumin from the crystalline state to the amorphous state. In the preparation of Cur/GANPs, the small size of the nanoparticles was due to the use of a high-pressure homogenizer (Fig. 2). Due to its negative charge, Cur/GANPs are useful for the effective delivery of curcumin as therapeutic agents. TEM image is depicted in Fig. 2b showing the semi-spherical shape of individual particles.

It can be suggested that the aggregation of Cur/GANPs in -20°C was due to the detachment of gum Arabic carriers from the nanoparticles prepared as showed by the decrease of zeta potential values. The detachments may be related to adjacent nanoparticles' interactions that cause the aggregation of Cur/GANPs suspensions. These findings are similar to previous studies, where larger particles are also produced by

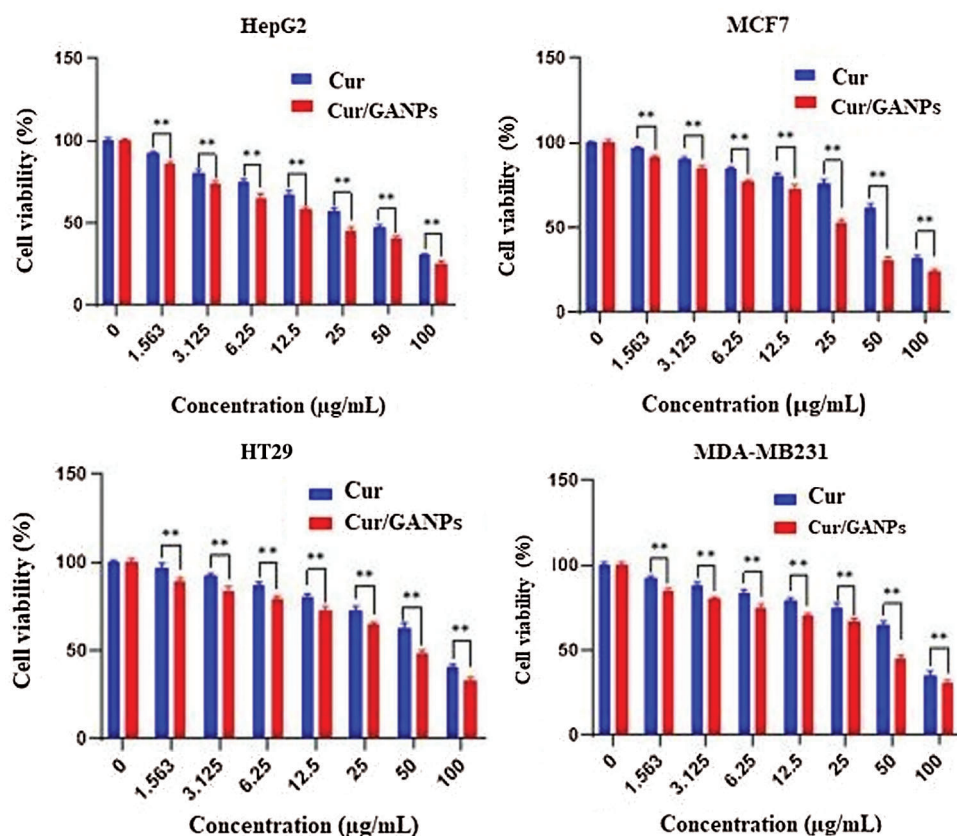


FIGURE 11. Cytotoxicity activity of curcumin, Cur/GANPs at different concentrations after 72 hrs of treatment in four human cancer cell lines A: HepG2, B: MCF-7, C: HT29, and D: MDA-MB231). Data shown are mean value \pm SD, independent *t*-test, * *p*-value \leq 0.05, ** *p*-value \leq 0.01.

TABLE 1

IC₅₀ of free curcumin, Cur/GANPs in HT29, MDA-MB231, MCF7, and HepG2 cell lines after 72 h of incubation. Each data represents the mean of three measurements \pm SD. *p* < 0.05 is considered significant

Treatment	IC ₅₀ (µg/mL)			
	HT29	MDA-MB231	MCF7	HepG2
Cu/GANPs	20.14 \pm 1.3	18.22 \pm 1.3	16.1 \pm 1.01	6.22 \pm 1.4
Free curcumin	36.9 \pm 1.5	33. \pm 1.4	30.1 \pm 1.1	15.4 \pm 1.2

the rapid freezing process due to the electrostatic repulsive and stronger van der Waals attractive forces that occurred between nanoparticles (Albanese *et al.*, 2012). Hence, the nanoparticles in powdered form are more stable than the preparation of suspensions. Also, the powdered nanoparticles improve the release of drugs in the dispersed phase with ease of storage during a long period. The use of cryoprotectant is recommended to decrease the risk of agglomeration of Cur/GANPs suspensions in temperature of storage of (4°C) and usage within 10 days. As shown in Fig. 4, the Zeta potential of Cur/GANPs decreased with the reduction of pH from 7.4 to 4.8, whereas the values of particle size of Cur/GANPs increased as the pH of solutions decreased to 4.8, mentioning the formation of larger particles due to the aggregation behavior caused by the weak electrostatic repulsion in the medium.

The release of Cur from GANPs was carried out using a phosphate buffer solution at pH 7.4 and pH 4.8. Fig. 5

exhibited that the release of Cur from GANPs at pH 4.8 was divided into two phases based on the slope of release profiles and release percentages. An initial release burst was recorded for the first 1000 min.

The rapid initial release of Cur at pH 7.4 and pH 4.8, respectively, was related to the weak absorption or bonding of the surface of Cur/GANPs. As shown in Fig. 5, curcumin was rapidly released from Cur/GANPs up to 65% and 33% at pH 4.8 and pH 7.4, respectively. A much slower release of Cur from Cur/GANPs in pH 7.4 over a prolonged period of 3000 min than pH 4.8.

The DPPH is characterized by its stability and its capacity to accept electrons or hydrogen radicals when it reacts with reducing agents to produce a diamagnetic hydrazine substance. In such a reaction the DPPH reagent loses its color depending on the amount of reduction. Thus any substance capable of donating electrons or hydrogen atoms in the reaction will lead to the conversion of purple DPPH to the yellow 1, 1-diphenyl-2- picrylhydrazine, a reaction that can be evaluated by measuring color intensity using the spectrophotometer with absorbance at 517 nm. As shown in Fig. 6, the Cur/GANPs had a higher percentage of antioxidant capacity than the pure curcumin which can be attributed to the reactivity of functional groups of gum Arabic. Antioxidant properties of gum Arabic (GA) has been established previously which confirms its biomedical applications as a nanocarrier for tumor-specific treatment (Gamal *et al.*, 2003).

Nitric oxide (NO) is an endogenous radical that is involved in regulating essential biological functions such as neurotransmission, controlling the vascular caliber, and blood flow. However, NO is also involved with pathological

events due to its contribution to oxidative damage. Nitric oxide (NO) plays a major role in inducing oxidative stress in several inflammatory processes. One of the reactive free radicals that is produced by the NO reaction with superoxide radical is peroxynitrite anion (ONOO⁻) which is a strong oxidant capable of interacting with organic molecules like lipids inducing the aberrant inflammatory responses (Mikkelsen and Wardman, 2003). The test aimed to determine the nitric oxide (NO) scavenging activity of the developed formulation upon stimulation of the macrophages with bacterial lipopolysaccharide (LPS). In this experiment, nitric oxide radicals are generated from sodium nitroprusside by the macrophages nitric oxide synthase enzyme that contributes with other immune mediators in the inflammatory responses (Dowding et al., 2012). Then the released NO radicals interacts with oxygen to produce nitrite ions in an aqueous solution at neutral pH which is measured by using the Griess reaction reagent. In the beginning, the potential toxicity of the macrophages in response to LPS stimulation was determined following treatment with curcumin and Cur/GANPs using the MTT assay. The same experiment setting was also repeated without LPS stimulation. As shown in Fig. 7a, the treatments had a minimal influence on the viability of RAW 264.7 cells until 500 µg/mL concentration. Furthermore, upon stimulation with LPS, the viability of the cells was less compared to the cells without stimulation; however, with Cur/GANPs nanoparticles treatment, the viability of the cells had significantly improved as shown in Fig. 7b. The results confirmed that both curcumin and Cur/GANPs had a minimal toxic effect on the murine macrophages with significantly less toxicity for the nanoformulation. This may be due to the biocompatibility of the carrier (GA) which was also demonstrated in other reports confirming that GA formulations had low toxicity in biological cells (Kong et al., 2014).

The mechanism of NO inhibition by curcumin and Cur/GANPs was not clarified in this study. In other reports, curcumin had also shown inhibitory activity on NO and other mediators of inflammation by inhibition of cyclooxygenase, nitric oxide synthase, lipoxygenase enzymes, and NF kappa B (Bhaumik et al., 2000; Bengmark, 2006).

The main concern in the determination of the antioxidant effects of different agents is due to their potential roles in protecting the biological system from the deleterious consequences of oxidizing agents including hydrogen peroxides and hydroxyl radicals (Castro et al., 2006). Like many extracts of traditional medicines, the therapeutic effects of Cur/GANPs may be attributed to its activity in scavenging free radicals. The reported therapeutic properties of curcumin were attributed to its activity in scavenging reactive oxygen species (Deguchi, 2015). The study demonstrated the potent antioxidant and anti-inflammatory properties of curcumin that were preserved and enhanced in the nanoformulation. This finding is consistent with other studies that confirmed GA's ability to preserve and enhance the antioxidant properties of phenolic compounds in various formulations (Pitalua et al., 2010; Ahmed et al., 2016). The enhanced antioxidant activity may be attributed to the antioxidant properties of GA that added a synergistic effect in the formulation. Gum Arabic had shown protective properties

against the attacks of lipoxygenase and free radicals (García-Martínez et al., 2018). In addition, GA had shown in studies its targeting specificity based on the interaction of hydroxyl groups with asialoglycoprotein receptors. Thus, this polymer has excellent potential for delivering bioactive agents as a target-oriented delivery system, functioning through receptor-mediated endocytosis (Gamal-Eldeen et al., 2016).

The β-carotene bleaching assay had been used to confirm the antioxidant properties of the NPs treatment. This assay depends on the oxidation of β-carotene by linoleic acid in the emulsion system causing rapid discoloration due to the formation of linoleate free radicals. Agents with the antioxidant effect would deter these changes and are measures using color changes.

β-carotene is a natural carotenoid derived from plants deemed as a major precursor of vitamin A. It functions as a free radical scavenger that protected the human body against the attack of free radicals (Yusuf et al., 2012). β-carotene is also reported to protect the nervous system from neurodegenerative disorders that are largely attributed to oxidative stress (Obulesu and Bramhachari, 2011). The principle of the β-carotene assay is based on linoleic acid oxidation, which resulted in a decrease of β-carotene's yellow color. This assay added another evidence to support the antioxidant potential of curcumin in the free and nanoparticle form that was demonstrated at various concentrations in comparison with α-tocopherol as shown in Fig. 9. The results showed that both curcumin and Cur/GANPs could inhibit ACE. In Fig. 10, after 90 min of incubation, 100 µL of 5 µg/mL of Cur/GANPs and free curcumin inhibited ACE with a maximal percentage of 72.3% and 55.4%, respectively.

The demonstrated antihypertensive effect of curcumin was consistent with findings in other studies; in which it had similar ACE inhibitory effects to the classical ACE inhibitor class of medications (Anand et al., 2007; Rachmawati et al., 2016; Shome et al., 2016). The angiotensin-converting-enzyme is one of the main regulatory elements in the cardiovascular system with therapeutic potentials being the target of several treatments. The ACE enzyme is a zinc metalloprotease a carboxypeptidase or dipeptidyl carboxypeptidase (EC 3.4.15.1) that is uniformly found in mammalian tissues with the main function of converting angiotensin I to the potent vasoconstrictor angiotensin II (Sarmadi et al., 2011). In this assay, the percentage inhibition of curcumin and Cur/GANPs was proportional to the decrease of absorbance measurements at 228 nm. Cur/GANPs had more potent ACE inhibitory effects compared to free curcumin, which may be attributed to the increase in bioavailability of curcumin and the synergistic properties of gum Arabic coating (Hassani et al., 2019; Jaafar, 2019). Therefore, it can be suggested that encapsulation of curcumin in gum Arabic could improve the antihypertensive capacity of curcumin and such a complexation of gum Arabic with curcumin may represent an important alternative antihypertensive treatment.

Apparently, free curcumin and Cur/GANPs had similar cytotoxicity in the cancer cell lines after 72 h of incubation. Such cytotoxic effects of curcumin were reported using nanoformulation in colon cancer cells *in vitro* without reporting the mechanism (Prajakta et al., 2009; Selvam et al., 2019). The

stipulated mechanism of curcumin anticancer properties was attributed in several studies to NF- κ B, an important transcription factor in the oncogenic signaling pathways in ovarian cancer cells *in vitro* and *in vivo* (Lin *et al.*, 2007), breast cancer cells (Chung and Vadgama, 2015). Although, other reports demonstrated curcumin anticancer properties that were attributed to interference with cell cycle, apoptosis and proliferation markers through modulation of cancer-specific micro RNAs in lung cancer (Lelli *et al.*, 2017) and pancreatic cancer cells (Nagaraju *et al.*, 2019).

Furthermore, the liver cancer cells (HepG2) were more sensitive to both treatments compared to other cancer cell lines with 50% of the cells inhibited (IC₅₀) by concentrations of 15.4 and 6.22 for curcumin and Cur/GANPs respectively (Table 1). This finding is consistent with a study conducted on HepG2 cells using free curcumin where significant inhibition of cell proliferation was detected and attributed to cell cycle and cytoskeletal interference (Jiang *et al.*, 2013). In this study, the cytotoxicity was significantly improved by the nanoformulation in which the anticancer activity of curcumin nanoparticles (Cur/GANPs) was two times higher than free curcumin. This may be attributed to the size of the delivered particles (Bioavailability) and the enhanced cellular uptake. In fact, some evidence pointed to the specific binding of the galactose groups of GA coating material with asialoglycoprotein receptors on the hepatocytes enhancing the cellular receptor-mediated endocytosis of curcumin (Sarika *et al.*, 2015). That specific binding enhanced the cytotoxicity in HepG2 cells compared with MCF-7 cells and such binding could partly explain the finding in this study. In addition to the unique properties of GA, particularly high stability and low toxicity.

Conclusion

Nano formulation of curcumin using Gum Arabic significantly enhances the antineoplastic properties of free curcumin. This adds further evidence to the therapeutic properties of curcumin in diseases like cancer and cardiovascular diseases due to its potency and lower toxicity compared with other chemotherapeutic agents. The effects are mainly attributed to the antioxidant properties that were further enhanced with Gum Arabic. These observations demand further investigations to uncover the underlying mechanisms and effects on the survival mechanisms of cancer cells such as invasion, metastasis, apoptosis, and other processes using *in vitro* and *in vivo* settings.

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Availability of Data and Materials: The datasets generated during and/or analyzed during the current study are available from the corresponding author on reasonable request.

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